

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

Double replacement/displacement

#### VI. Heating Cu with Atmospheric O<sub>2</sub>

appearance the unreacted Cu mesh:

shiny copper

appearance of the Cu mesh after heating and cooling:

dull dark brown

chemical equation:



general reaction type:

combination

Which species are oxidized?

Cu

Which species are reduced?

O

#### VII. Reacting CuSO<sub>4</sub> Solution with Steel Wool (Fe)

appearance of the steel wool (Fe) before reaction:

silver solid

appearance of the CuSO<sub>4</sub> solution before reaction:

light blue

appearance of the steel wool (Fe) after the reaction has occurred:

dark brown solid

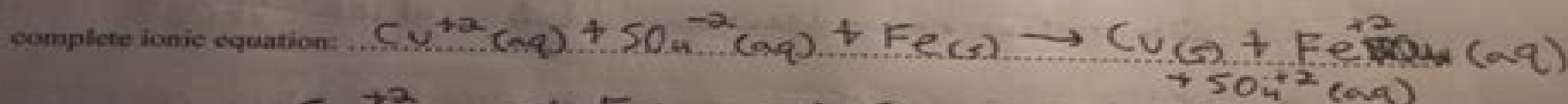
appearance of the CuSO<sub>4</sub> solution after the reaction has occurred:

light blue liquid

molecular equation:



complete ionic equation:



net ionic equation:



spectator ions:



general reaction type:

single displacement

Which species are oxidized?

Fe

Which species are reduced?

Cu<sup>+2</sup>

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**James F. Hall**



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web reading comprehension 300 level primary 6 minute solution questions these 25 pages of questions and 25 pages of  
answers were developed in order to work on