## Chapter 7: Chemical Reactions

A chemical reaction: a process in which at least one new substance is formed as the result of a chemical change.

Evidence that a chemical reaction has occurred:

- Color change
- > Formation of a solid (in a previously clear solution)
- Formation of a gas
- Emission of light
- Emission or absorption of heat

A chemical equation: a written statement the uses symbols and formulas instead of words to describe the changes that occur in a chemical reaction

Ex.

$$MgO + C \rightarrow CO + Mg$$

Special symbols are often used in chemical equations to express the state of reactants and products

Ex.

$$Mg(OH)_{2(s)} + 2 HCI_{(aq)} \rightarrow MgCI_{2(aq)} + 2 H_2O_{(l)}$$

# **Chemical Reactions Chapter 7 Test A Answers**

Carla C. Kirkland, Chan Cleveland

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