

# Chemical Reactions

## Signs of Chemical Reactions

What is a chemical reaction? \_\_\_\_\_

## Chemical Formulas

A chemical formula uses \_\_\_\_\_ to identify the \_\_\_\_\_ in a compound and the \_\_\_\_\_ of each kind present in a molecule.



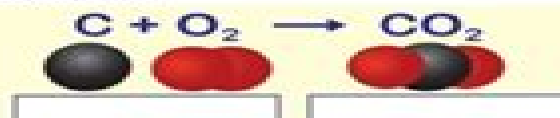
A water molecule is made up of \_\_\_\_\_ atoms of \_\_\_\_\_ bonded to \_\_\_\_\_ atom of \_\_\_\_\_.



A glucose molecule is made up of \_\_\_\_\_ atoms of \_\_\_\_\_, \_\_\_\_\_ atoms of \_\_\_\_\_, and \_\_\_\_\_ atoms of \_\_\_\_\_.

## Chemical Equations

A chemical equation uses \_\_\_\_\_ and \_\_\_\_\_ to describe a chemical reaction. The **starting materials** in a chemical reaction are \_\_\_\_\_. The **resulting substances** from a reaction are \_\_\_\_\_. The \_\_\_\_\_ are always placed on the left of the arrow, and the \_\_\_\_\_ on the right of the arrow.



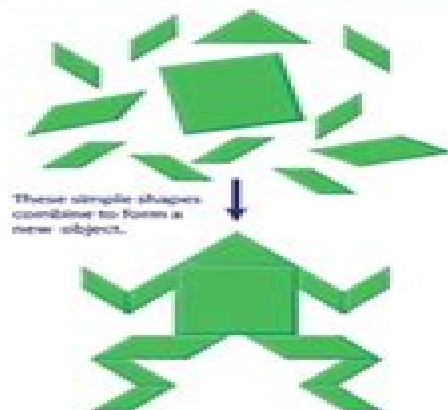
## Key Vocabulary Terms

- atom
- chemical equation
- chemical formula
- chemical reaction
- compound
- decomposition reaction
- double replacement reaction
- element
- product
- reactant
- single replacement reaction
- synthesis reaction

## Types of Chemical Reactions

### Synthesis Reaction

Describe a synthesis reaction:



### Decomposition Reaction

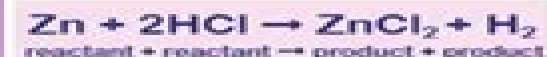
Describe a decomposition reaction:



### Replacement Reaction

#### Single Replacement

Describe a single replacement reaction:



#### Double Replacement

Describe a double replacement reaction:



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# Chemical Reactions Guide

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