# Chemical Reactions

# Signs of Chemical Reactions

What is a chemical reaction?

#### Chemical Formulas

A chamical formula uses \_ too lidentify the in a compound and the

H<sub>2</sub>O water A worther excelerously is enader up-of \_\_\_\_\_ atoms of

bonded to \_\_\_\_\_ atom of



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			.and.		_atoms of	

## **Chemical Equations**

A chemical equation uses to describe a

chemical reaction. The starting materials in a chemical reaction are . The resulting substances from a reaction

placed on the left of the amow, and the the right of the arrow.

C + O. - CO.

#### **Key Vocabulary Terms**

- double replacement reaction.
- chemical equation. chemical formula
- chemical reaction.
  - \* PRINCESOR
- \* compound decomposition reaction - synthesis reaction.

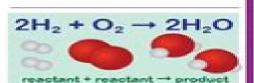
\* autopern

- \* effectivent \* percellant
- single replacement reaction

Types of Chemical Reactions

#### Synthesis Reaction

Describe a synthesis reaction:





contribution top Foreign in mere adopted.

#### Decomposition Reaction

Describe a decomposition reaction:

2H	2O -	+ 2H <sub>2</sub>	+ 02
4	0	0	1
		00	
Fee	ectant	product +	product



#### Replacement Reaction

#### Single Replacement

Describe a single replacement reaction:

Zn + 2HCl → ZnCl<sub>2</sub>+ H<sub>2</sub> reactant + reactant -+ product + product



#### Double Replacement

Describe a double replacement reaction:

NaCl + AgF → NaF + AgCl reactant + reactant -+ product + product



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