

**Unit 11: Chapter 7 - Chemical Reactions Review Answer Key**

**Define:**

1. **Reactant:** *A substance that undergoes change in a chemical reaction.*
2. **Product:** *New substances formed as a result of a chemical reaction.*
3. **Catalyst:** *A substance that affects the rate of a chemical reaction without being used up in the reaction.*
4. **Coefficient:** *Numbers that appear before a formula in a chemical equation to show the relative proportions of each reactant and product.*
5. **Subscript:** *A number written lower in a chemical formula indicating the amount of atoms present of an element (this number is not to be changed once written in chemical formula).*
6. **Law of Conservation of Mass:** *Matter cannot be created nor destroyed; only changed (atoms can be rearranged).*
7. **Synthesis reaction (include an example):** *A chemical reaction in which two or more substances react to form a single substance.*  $(A + B \rightarrow AB)$
8. **Decomposition reaction (include an example):** *A chemical reaction in which a compound breaks down into two or more simpler substances.*  $(AB \rightarrow A + B)$
9. **Single replacement (include an example):** *A chemical reaction in which one element takes the place of another element in a compound.*  $(AB + C \rightarrow AC + B)$
10. **Double replacement (include an example):** *A chemical reaction in which two compounds exchange positive ions and form two new compounds.*  $(AB + CD \rightarrow AD + CB)$
11. **Combustion (include an example):** *A chemical reaction in which a substance reacts rapidly with oxygen, often producing heat and light.*  $(C_2H_6 + O_2 \rightarrow CO_2 + H_2O)$
12. **Endothermic reaction:** *A chemical reaction that absorbs energy from its surroundings.*
13. **Exothermic reaction:** *A chemical reaction that releases energy to its surroundings.*
14. **Chemical equation (include an example):** *A representation of a chemical reaction in which the reactants and products are expressed as formulas.*
15. A **balanced equation** is an equation in which the number of atoms on the reactants' side **equals** the number of atoms on the products' side of the chemical equation.
16. **Equilibrium:** *A state in which the forward and reverse paths of a physical or chemical change take place at the same rate.*

# Chemical Reactions Review And Reinforce Answers

**Dr. Ashokan Kannarath**



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Foundation,2012-04-11 CK 12 Biology Teacher s Edition complements the CK 12 Biology Student Edition FlexBook

**Matter** Prentice-Hall Staff,1994      **Prentice Hall Science Explorer** Michael J. Padilla,2002      Prentice Hall

Chemistry ,2000      **Prentice Hall Exploring Life Science** Anthea Maton,1997      Mastering NEET MCQs: Speed-solving Techniques for Medical Aspirants Dr. Ashokan Kannarath,2024-01-31 This book dives deep into the strategies techniques and tips required to excel in the NEET MCQs It is designed to help you not only solve questions quickly but also enhance your understanding and retention of vital medical concepts It explains how to analyse NEET MCQs without getting overwhelmed efficiently identifying keywords understanding question patterns and eliminating wrong answer options It also exposes proven strategies to increase your answering speed without compromising accuracy Includes time management tips mental math tricks and effective test taking methods      *Focus on Earth Science* ,2001      **Chemical Interactions** ,2005      Earth Science ,2001      *Human Biology and Health* ,1997      Cells ,1997      **Paint, Oil and Chemical Review ...** ,1915

Fundamentals of Modern Manufacturing Mikell P. Groover,2010-01-07 Engineers rely on Groover because of the book s quantitative and engineering oriented approach that provides more equations and numerical problem exercises The fourth edition introduces more modern topics including new materials processes and systems End of chapter problems are also thoroughly revised to make the material more relevant Several figures have been enhanced to significantly improve the quality of artwork All of these changes will help engineers better understand the topic and how to apply it in the field      An Introduction to Physical Science James T. Shipman,Jerry D. Wilson,Aaron W. Todd,2002-07-08 This text presents a survey of the physical sciences physics chemistry astronomy meteorology and geology for non science majors Topics are treated both descriptively and quantitatively providing flexibility for instructors who wish to emphasize a highly descriptive approach a highly quantitative approach or any spectrum in between      *Human Biology and Health* Michael J. Padilla,2002      **Life Science, Grades 6-7** Tony Wright,1994-07-13      Physical Science Anted, Grades 6-9 David W. Appenbrink,1986      **The Software Encyclopedia 2001** ,2001      **Ess Chem Probs Study Guide** David Margolese,David W. Oxtoby,2004-04

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