

## Balancing & Reactions Test SCH3U

Name: \_\_\_\_\_

    K       T       C       A   Assessment of Learning (Product)  
**10      20      6      7**

Multiple Choice: Circle the most correct answer. (10K)

1. Classify the following chemical reaction:  $2\text{AsCl}_3 + 3\text{H}_2\text{S} \rightarrow \text{As}_2\text{S}_3 + 6\text{HCl}$

- A) combustion                      B) synthesis                      C) decomposition  
D) single displacement           E) double displacement

2. The type of reaction that takes place when one element reacts with a compound to form a new compound and a different element is a \_\_\_\_\_.

- A) synthesis reaction                      C) single-displacement reaction  
B) decomposition reaction                D) double-displacement reaction

3. Which of the following types of chemical reaction produces water and carbon dioxide gas?

- A) single displacement                      B) double displacement           C) decomposition           D) combustion

4. When the equation  $\text{NH}_4\text{OH} + \text{H}_2\text{SO}_4 \rightarrow (\text{NH}_4)_2\text{SO}_4 + \text{H}_2\text{O}$  is balanced, one term is:

- A)  $\text{NH}_4\text{SO}_4$     B) 2  $\text{H}_2\text{O}$            C)  $\text{H}_2\text{OH}$            D) 2  $(\text{NH}_4)_2\text{SO}_4$

5. When the eq.  $\text{H}_2 + \text{V}_2\text{O}_5 \rightarrow \text{V} + \text{H}_2\text{O}$  is completed and balanced, one term is:

- A) 2  $\text{V}_2\text{O}_5$     B) 3  $\text{H}_2\text{O}$            C) 2 V           D) 8  $\text{H}_2$

6. Boron trifluoride reacts with water to produce two products,  $\text{H}_3\text{BO}_3$  and  $\text{HBF}_4$ . Which of the following equations correctly represents this reaction?

- A)  $3\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + \text{HBF}_4$   
B)  $2\text{BF}_3 + 2\text{H}_2\text{O} \rightarrow 2\text{H}_3\text{BO}_3 + \text{HBF}_4$   
C)  $4\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + 3\text{HBF}_4$   
D)  $4\text{BF}_3 + 3\text{H}_2\text{O} \rightarrow \text{H}_3\text{BO}_3 + 3\text{HBF}_4$

7. In a double-displacement reaction, the \_\_\_\_\_.

- A) products are always molecular           B) reactants are usually two ionic compounds  
C) reactants are two elements                D) products are a new element and a new compound

8. Which of the following statements is true about single-displacement reactions?

- A) They are restricted to metals.                C) Two reactants produce two products.  
B) They involve a single product.              D) Any metal replaces any other metal.

9. Some acids, such as carbonic acid, decompose to nonmetal oxides and

- A) water           B) a salt           C) oxygen           D) peroxide

# Chemical Reactions Test Answers

**Yijin Wang**



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