

____K

Assessment of Learning (Product)

10

Multiple Choice: Circle the most correct answer. (10K)

1. Given the following reaction: $\text{H}_2(\text{g}) + \text{Cl}_2(\text{g}) \rightarrow 2 \text{HCl}$ As this reaction takes place:

- A) H_2 and Cl_2 are consumed at the same rate
- B) HCl is produced at the same rate that H_2 is consumed.
- C) HCl is produced at the half the rate that Cl_2 is consumed
- D) HCl is produced at the same rate that Cl_2 is consumed

2. The minimum amount of energy required to overcome the energy barrier in a chemical reaction is the:

- A) heat of reaction
- B) activation energy
- C) KE of the reactants
- D) enthalpy of the products

3. The average kinetic energy of colliding particles can be increased by:

- A) adding a catalyst.
- B) increasing pressure
- C) increasing temperature
- D) increasing reactant concentration.

4. Consider the following reaction: $2\text{H}_2\text{O}_{(g)} \rightarrow 2\text{H}_2\text{O}_{(l)} + \text{O}_{2(g)}$

Which graph shows the relationship between rate of consumption of H_2O_g and time?

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Industrial & Engineering Chemistry ,1925

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