

1

Sample Paper

ICSE Specimen Paper

17th June, 2023

Time Allowed : 2 Hours

CHEMISTRY
(Science Paper-2)

Maximum Marks : 80

General Instructions:

- (1) Answer to this paper must be written on the paper provided separately.
- (2) You will not be allowed to write during first 15 minutes.
- (3) This time is to be spent in reading the question paper.
- (4) The time given at the head of this paper is the time allowed for writing the answers.
- (5) Section A is compulsory. Attempt any four questions from Section B.
- (6) The intended marks for questions or part of questions are given in brackets [].

SECTION - A

[40 Marks]

(Attempt all questions from this Section.)

1. Choose the correct answers to the questions from the given options. [15]

(Do not copy the questions, write the correct answers only.)

- (i) An aqueous solution of copper sulphate turns colourless on electrolysis. Which of the following could be the electrodes?
P. anode: copper; cathode: copper
Q. anode: platinum; cathode: copper
R. anode: copper; cathode: platinum
(a) only P
(b) only Q
(c) only R
(d) both Q and R
- (ii) A compound P is heated in a test tube with sodium hydroxide solution. A red litmus paper held at the mouth of the test tube turns blue. Which of the following could compound P be?
(a) zinc sulphate
(b) copper sulphate
(c) ferrous sulphate
(d) ammonium sulphate

- (iii) The atomic masses of sulphur (S), oxygen (O), and helium (He) are approximately 32, 16, and 4 respectively.

Which of the following statements regarding the number of atoms in 32 g of sulphur, 16 g of oxygen, and 4 g of helium is correct?

- P. 16 g of oxygen contains four times the number of atoms as 4 g of helium.
Q. 16 g of oxygen contains half the number of atoms as 32 g of sulphur.
(a) only P
(b) only Q
(c) both P and Q
(d) neither P nor Q

- (iv) Ammonia gas is passed through quicklime and then collected in a jar. Red and blue litmus papers are placed in the jar. W, X, Y and Z are the four observations.

Which of the above observations correctly shows the reaction of the litmus papers to ammonia?

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Christian G. Meyer



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