

# CHEMISTRY MULTIPLE CHOICE QUESTIONS

## A. Atoms, molecules and stoichiometry

2002 -2019



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C.

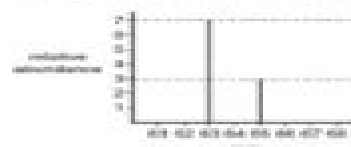
As a simplification, an adult human can be considered to have a daily diet of 1.80 kg of carbohydrate (empirical formula  $C_6H_{12}O_6$ ).

Which mass of carbon dioxide does a person produce each day if all the carbohydrate eaten is digested and oxidised?

- A. 0.907 kg    B. 0.900 kg    C. 1.30 kg    D. 2.54 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A. 63.2    B. 63.5    C. 63.8    D. 64.0

(2002 NA/10)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 MC/31)

A mixture of 1.0 cm<sup>3</sup> of methane and 1.0 cm<sup>3</sup> of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula  $CH_4O$ ?

- methanol
- ethanoic acid
- methyl methanolate

(2002 ON/31)

What is the number of molecules in 500 cm<sup>3</sup> of oxygen under room conditions?

- A.  $1.25 \times 10^{23}$     B.  $1.25 \times 10^{24}$     C.  $3.0 \times 10^{23}$     D.  $3.0 \times 10^{24}$

(2008 MC/12)

In the preparation of soft margarine, glycerol trioleate reacts



is fully hydrogenated so that, on average, one of its side-chains is converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2(\text{CH}_2)_7\text{CO}_2$  moiety and two side-chains are converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2(\text{CH}_2)_7\text{CO}_2$  moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A. 4    B. 5    C. 6    D. 8

(2008 MC/22)

Use of the Data Booklet is relevant to this question.

Additional classes are about very small amounts of water which dissolve in 1000 and

# Chemistry 20multiple Choice Questions

**John T. Moore, Richard H. Langley**



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Synopsis of Chemistry with Question Bank & Mnemonics Dr Gurcharan Dass, Dr K Chaudhry, Synopsis of Chemistry is intended to help NEET UG candidates score optimum results Based upon CBSE syllabus the book is useful for Class 11 and Class 12 Students Dr Gurcharan Dass is Principal of Manohar Memorial Postgraduate College Fatehabad He has huge teaching experience Dr K Chaudhry was First Author of Jaypee Brothers He developed FIRST 1 ml Colorimeter in India He is Freelance Teacher on Youtube **A Test of High-school Chemistry** Henry Lester Gerry, 1924 CliffsNotes AP

Chemistry Bobrow Test Preparation Services, 2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter's end Two full length practice tests with detailed answer explanations are included in the book **Getting Into University:**

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