

## ***CHEM 111 Study Guide for Exam 4: Chapters 7–9***

### **Chapter 7: Chemical reactions and chemical quantities**

1. Know how to do gram / mole conversions.
2. Know the relationships between moles and chemical equations, and how to do calculations based on a balanced equation
3. Know the relationships between mass and chemical equations, and how to do calculations based on a balanced equation
4. Be able to figure out limiting reagent and be able to calculate theoretical yield and percent yield.

### **Chapter 8: Introduction to Solutions and Aqueous Reactions**

1. Understand molarity as a concentration and be able to use it in calculations.
2. Be able to do dilution stoichiometry.
3. Know the difference among strong electrolytes, weak electrolytes and nonelectrolytes.
4. Be able to do solution stoichiometry in a given chemical reaction.
5. Be able to write molecular equations, complete ionic equations and net ionic equations.
6. Be able to assign oxidation numbers (states) for a given chemical formula.
7. Be able to recognize redox reactions, and identify oxidation reaction, reduction reaction, oxidizing agent and reducing agent.

### **Chapter 9: Thermochemistry**

1. Know the difference of kinetic energy and potential energy.
2. Understand endothermic reactions and exothermic reactions.
3. Be able to write equations for standard enthalpy of formation ( $\Delta H_f^\circ$ ).
4. Be able to use Hess's law.
5. Be able to use bond energies to calculate  $\Delta H$  for a chemical reaction.
6. Be able to use given  $\Delta H_f^\circ$  to calculate  $\Delta H_{\text{rxn}}^\circ$
7. Be able to calculate heat (q) using specific heat:  $q = c.m. \Delta T$

# Chemistry 111 Study Guide

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