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ICE3.1. What volume of 0.778 M sodium carbonate solution must be diluted to 150.0 mLs with water to reduce its concentration to 0.0234 M Na_2CO_3 ?

$$M_C \cdot V_C = M_D \cdot V_D$$

$$0.778 \text{ M} \cdot V_C = 0.0234 \text{ M} \cdot 150.0 \text{ mL}$$

$$V_C = \frac{0.0234 \text{ M} \cdot 150.0 \text{ mL}}{0.778 \text{ M}} \\ = 4.51 \text{ mL}$$

ICE3.2. A 0.0945 g sample of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ is dissolved and diluted to the mark in a 500.0 mL volumetric flask. A 2.00 mL sample of this solution is transferred to a second 5000.0 mL volumetric flask and diluted.

a) What is the molarity of the CuSO_4 in the final solution?

$$0.0945 \text{ g CuSO}_4 \cdot 5\text{H}_2\text{O} \left(\frac{1 \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}}{250. \text{ g}} \right) = 3.78 \times 10^{-4} \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O} \\ \frac{3.78 \times 10^{-4} \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}}{0.500 \text{ L}} = 7.56 \times 10^{-4} \text{ M CuSO}_4$$

$$M_C \cdot V_C = M_D \cdot V_D$$

$$7.56 \times 10^{-4} \text{ M CuSO}_4 \cdot 2.00 \text{ mLs} = M_D \cdot 5000.0 \text{ mL}$$

$$M_D = \frac{7.56 \times 10^{-4} \text{ M CuSO}_4 \cdot 2.00 \text{ mLs}}{5000.0 \text{ mLs}} \\ = 3.02 \times 10^{-7} \text{ M CuSO}_4$$

I meant to a second 500.0 mL volumetric flask

$$M_C \cdot V_C = M_D \cdot V_D$$

$$7.56 \times 10^{-4} \text{ M CuSO}_4 \cdot 2.00 \text{ mLs} = M_D \cdot 500.0 \text{ mL}$$

$$M_D = \frac{7.56 \times 10^{-4} \text{ M CuSO}_4 \cdot 2.00 \text{ mLs}}{500.0 \text{ mLs}} \\ = 3.02 \times 10^{-6} \text{ M CuSO}_4$$

b) To prepare the solution directly what mass of $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ must be weighed out?

$$5.000 \text{ L} \left(\frac{3.02 \times 10^{-6} \text{ M CuSO}_4}{1 \text{ L}} \right) = 1.51 \times 10^{-6} \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}$$

$$1.51 \times 10^{-6} \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O} \left(\frac{250. \text{ g}}{1 \text{ mol CuSO}_4 \cdot 5\text{H}_2\text{O}} \right) = 3.78 \times 10^{-4} \text{ g CuSO}_4 \cdot 5\text{H}_2\text{O}$$

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Chao Zhang



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Table of Contents Chemistry 1314 Laboratory Manual Key

1. Understanding the eBook Chemistry 1314 Laboratory Manual Key
 - The Rise of Digital Reading Chemistry 1314 Laboratory Manual Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 1314 Laboratory Manual Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 1314 Laboratory Manual Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 1314 Laboratory Manual Key
 - Personalized Recommendations
 - Chemistry 1314 Laboratory Manual Key User Reviews and Ratings
 - Chemistry 1314 Laboratory Manual Key and Bestseller Lists
5. Accessing Chemistry 1314 Laboratory Manual Key Free and Paid eBooks

- Chemistry 1314 Laboratory Manual Key Public Domain eBooks
- Chemistry 1314 Laboratory Manual Key eBook Subscription Services
- Chemistry 1314 Laboratory Manual Key Budget-Friendly Options
- 6. Navigating Chemistry 1314 Laboratory Manual Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemistry 1314 Laboratory Manual Key Compatibility with Devices
 - Chemistry 1314 Laboratory Manual Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry 1314 Laboratory Manual Key
 - Highlighting and Note-Taking Chemistry 1314 Laboratory Manual Key
 - Interactive Elements Chemistry 1314 Laboratory Manual Key
- 8. Staying Engaged with Chemistry 1314 Laboratory Manual Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 1314 Laboratory Manual Key
- 9. Balancing eBooks and Physical Books Chemistry 1314 Laboratory Manual Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 1314 Laboratory Manual Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry 1314 Laboratory Manual Key
 - Setting Reading Goals Chemistry 1314 Laboratory Manual Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry 1314 Laboratory Manual Key
 - Fact-Checking eBook Content of Chemistry 1314 Laboratory Manual Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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