

PRACTICE EXAM 2

CHEM 1411 (Chap. 1, 2, 3, 4, 5)

- d 1. Which of the following is a physical change?
 - (a) rusting a nail
 - (b) fruits ripen
 - (c) baking a cake
 - (d) ice melts
 - (e) milk turns sour
- e 2. Which of the following is a homogeneous mixture?
 - (a) water and sand
 - (b) sand, alcohol, and sugar
 - (c) oil and vinegar
 - (d) milk of magnesium
 - (e) air
- e 3. Which of the following is a correct?
 - (a) sodium, S
 - (b) mercury, Mg
 - (c) tin, N, Sn
 - (d) potassium, P
 - (e) gold, Au
- b 4. What is the equivalent temperature for 105 Fahrenheit in Kelvin?
 - (a) 41
 - (b) 314
 - (c) 378
 - (d) 457
 - (e) -32
- a 5. A sample of carbon tetrachloride has a mass of 39.75 g and a volume of 25.0 mL. What is its density (g/mL)?

$d = \frac{m}{V} = \frac{39.75}{25.0 \text{ mL}}$

 - (a) 1.59
 - (b) 1.74
 - (c) 2.36
 - (d) 2.89
 - (e) 3.50
- e 6. How many significant figures are there in the measurement of 1.0356×10^4 ?
 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- c 7. How many significant figures are appropriate to show in the result after carrying out the operation below?

$(13.7 + 0.027) + 8.221 = ?$ \times, \div least s.f. $+, -$ least decimal

13.727

 - (a) 1
 - (b) 2
 - (c) 3
 - (d) 4
 - (e) 5
- d 8. How many protons, electrons, and neutrons are there in $^{130}\text{Ba}^{2+}$?
 - (a) 56p, 56e, 130n
 - (b) 56p, 54e, 130n
 - (c) 56p, 58e, 74n
 - (d) 56p, 54e, 74n
 - (e) 54p, 56e, 74n
- c 9. Which of the following pairs would have similar chemical and physical properties?
 - (a) Na, Mg
 - (b) H, Li
 - (c) He, Ne
 - (d) Cr, Ca
 - (e) Cl, O
- c 10. Which of the following is the molecular formula?
 - (a) NaCl
 - (b) C_2F_4
 - (c) HCl
 - (d) AlBr_3
 - (e) NaCN
- e 11. Which of the following is the empirical formula?
 - (a) C_2H_4
 - (b) CH_3COOH
 - (c) P_2O_{10}
 - (d) NH_4NO_3
 - (e) Ag_2CO_3
- e 12. Which of the following is a correct match?
 - (a) BaCl_2 , barium dichloride
 - (b) CS_2 , carbon sulfide
 - (c) CuF_2 , copper (I) fluoride
 - (d) HNO_2 , nitrous acid
 - (e) $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$, magnesium sulfate heptahydrate
- b 13. Which of the following pairs is a correct match?
 - (a) $\text{Ca}_3(\text{PO}_4)_2$, calcium (II) phosphate
 - (c) Mg_2O_3 , magnesium oxide
 - (b) $(\text{NH}_4)_2\text{SO}_4$, ammonium sulfate
 - (e) H_2CO_3 , carbon acid
 - (d) $\text{Cu}(\text{NO}_3)_2$, copper dinitrate

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Synchrotron Radiation C. Kunz,2013-11-11 Synchrotron radiation as a spectroscopic research tool has undergone a most inter esting and astonishing historical development and has now come to the stage of an exciting boom The machines which produce synchrotron radiation were built and de veloped exclusively for other purposes in the past namely high energy physics At the same time however they involuntarily became better and better light sources for the spectral range from the visible to the hard x ray region Now we are at the point that the first few storage rings have gone into operation as machines dedicated to synchrotron radiation and several more are in the stage of construction and planning All this was brought about by the successful research performed during the past fifteen years in which several groups allover the world haVe participated at dif ferent accelerator centers mostly symbiotic with high energy physics As it happens with a young and rapidly developing field the number of reviews and monographs is still minute The objective of this book is to fill an apparent gap and to provide a sound basis for those who are interested in synchrotron radiation and its applica tions

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