



HOUSTON COMMUNITY COLLEGE SYSTEM

Page 1

General Chemistry 1412 Practice Exam 1

Name: _____

Section: 1412-00000

Part 1: Answer 1-10. All 10 are 5 points each unless otherwise indicated. All 10 words/phrases
must be used in your answer.

1. Calculate the molarity of a solution (prepared by dissolving 1.0 g of water, H_2O , 1.000 g, in 1.000 g of H_2O).

A. 0.010 M

B. 0.010 M

C. 1.010 M

D. 0.0100 M

E. 0.0100 M

2. Which of the following will be most soluble in H_2O ?

A. CH_4

B. H_2

C. CH_3OH

D. C_2H_6

E. NH_3

3. Calculate the molarity of a solution (prepared by dissolving 1.0 g of water, H_2O , 1.000 g, in 1.000 g of H_2O). This depends on the definition of H_2O .

A. 0.0100 M

B. 0.010 M

C. 1.010 M

Chemistry 1412 Exams With Answers

RJ Alexander



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