

# Chemistry Worksheet



Calculate the molar mass of the following compounds:

- 1.  $\text{H}_2\text{O}$
- 2.  $\text{CO}_2$
- 3.  $\text{CH}_4$
- 4.  $\text{NH}_3$
- 5.  $\text{HCl}$
- 6.  $\text{NaCl}$
- 7.  $\text{KCl}$
- 8.  $\text{CaCl}_2$
- 9.  $\text{MgCl}_2$
- 10.  $\text{AlCl}_3$

ANSWER SHEET

- 1.  $\text{H}_2\text{O}$  18 g/mol
- 2.  $\text{CO}_2$  44 g/mol
- 3.  $\text{CH}_4$  16 g/mol
- 4.  $\text{NH}_3$  17 g/mol
- 5.  $\text{HCl}$  36.5 g/mol
- 6.  $\text{NaCl}$  58.5 g/mol
- 7.  $\text{KCl}$  74.5 g/mol
- 8.  $\text{CaCl}_2$  111 g/mol
- 9.  $\text{MgCl}_2$  95 g/mol
- 10.  $\text{AlCl}_3$  133.5 g/mol

## Calculating Molar Mass of Compounds

# Chemistry 1 Calculating Formula Mass Answer Key

**Peter Atkins, Julio de Paula**



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