

CHAPTER 8 REVIEW

Chemical Equations and Reactions**SECTION 1: Describing Chemical reactions**

1. Match the symbol on the left with its appropriate description on the right.



(a) A precipitate forms.



(b) A gas forms.



(c) A reversible reaction occurs.



(d) Heat is applied to the reactants.



(e) A chemical is dissolved in water.



(f) A chemical is in the liquid state.

2. Finish balancing the following equation:



3. In each of the following formulas, write the total number of atoms present.



4. Convert the following word equation into a balanced chemical equation: aluminum metal + copper(II) fluoride
- \rightarrow
- aluminum fluoride + copper metal



5. One way to test the salinity of a water sample is to add a few drops of silver nitrate solution with a known concentration. As the solutions of sodium chloride and silver nitrate mix, a precipitate of silver chloride forms, and sodium nitrate is left in solution. Translate these sentences into a balanced chemical equation.



Chemistry 1describing Chemical Reaction Section Review Answers

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Holt Chemistry Ralph Thomas Myers,2004 **Modern Chemistry** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 Glencoe Science McGraw-Hill Staff,2001-06 **Let's Review** Albert S. Tarendash,1993-09-01 Covers phases of matter atomic structure the chemical bond the periodic table solutions chemical reactions equilibrium acids and bases organic chemistry and lab procedures Chemistry Ken Hughes,Paul B. Kelter,1998-08 **Brief Review in Chemistry** Patrick Kavanah,2003

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