

	A		F	G	B	C
D	E		H	J	K	L
M			N	P		
				Q		

(a) Element Q belongs to period 5 and group VI. Place the element in the correct position in Figure 1. (1 mark)

(b) Consider the following ions:  $J^{2+}$ ,  $K^{-}$  and  $M^{+}$ .

(i) Write the electron arrangements for each. (2 marks)

2. 8. 8.

2-8-8

2. 8. 2

$$(S^{2-})$$

2-

(ii)  $\text{I}^{2-}$  It has the lowest nuclear charge / effective nuclear charge / Lowest no of proton.

(2 marks)



# Chemistry 20011 Kcse Paper 2

**Jenny Sharwood**



## Chemistry 20011 Kcse Paper 2:

Peak Revision K.C.S.E. Chemistry , Papers for Section 2, Chemistry ANZAAS (Association). Congress,1976 2 Unit  
Chemistry Higher School Certificate Examination Paper and Answers, 1999 Catherine Odlum,2000-01-01 SIMPLE  
TRICKS & TIPS IN CHEMISTRY PART-2 (For Class 12) Dr. Monica Bhandari,2021-11-24 A STRATEGIC BOOK THAT GIVES  
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management skills However several students generally do not get the expected rank score despite knowing all the concepts  
It is an irony that students spend several precious hours and parents spend huge money running from one coaching institute  
to another but nobody guides them on time management right from the beginning and when students attend crash courses  
about 1 2 months before competitive exam they hardly have any time to master these skills The author of this book strongly  
believes that with proper and timely strategic guidance on time management every student can achieve well in Competitive  
Exams This book has the strength to cover the entire syllabus in 10 days For fast track coverage of the entire course online  
coaching students can contact the author at monicabhandari books gmail com This book encompasses 10 chapters Each  
chapter starts with a quick review of the important concepts followed by shortcuts tips tricks The strategic manner in which  
topics are inter linked in this book will help in building a strong conceptual clarity among the students For illustration  
questions from previous years papers of JEE and NEET with accurate shortest possible solutions are provided along with  
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here s the last but not the least and the most important tip revise all your incorrectly attempted questions repeatedly and  
bookmark important questions so that you gain perfection in attempting questions having similar concept or trick IT S NOT  
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**Sqa** Leckie & Leckie,Scottish Qualifications Authority,2006-09-01 **Practical/Laboratory Manual Chemistry Class - XI**  
Er. Meera Goyal,2021-05-29 1 Basic Laboratory Techniques 1 To cut a glass tube or glass rod 2 To bend the glass rod at an  
angle 3 To draw a glass jet from a glass tube 4 To bore a cork and fit a glass tube into it Viva Voce 2 Characterisation and  
Purification of Chemical Substances 1 To determine the melting point of the given unknown organic compound and its  
identification simple laboratory technique Viva Voce 2 To determine the boiling point of a given liquid when available in small  
quantity simple laboratory method Viva Voce 3 To prepare crystals of pure potash alum  $K_2SO_4 \cdot Al_2(SO_4)_3 \cdot 24H_2O$  from the  
given impure sample 4 To prepare the pure crystals of copper sulphate from the given crude sample 5 To prepare pure

crystals of benzoic acid from a given impure sample Viva Voce 3 Measurement of pH Values 1 To determine the pH value of vegetable juices fruit juices tap water and washing soda by using universal pH paper 2 To determine and compare the pH values of solutions of strong acid HCl and weak acid  $\text{CH}_3\text{COOH}$  of same concentration 3 To study the pH change in the titration of strong base Vs strong acid by using universal indicator paper 4 To study the pH change by common ion  $\text{CH}_3\text{COO}^-$  ion in case of weak acid  $\text{CH}_3\text{COOH}$  5 To determine the change in pH value of weak base  $\text{NH}_4\text{OH}$  in presence of a common ion  $\text{NH}_4^+$  Viva Voce 4 Chemical Equilibrium 1 To study the shift in equilibrium between ferric ions and thiocyanate ions by changing the concentrations of either of the ions 2 To study the shift in equilibrium between  $\text{Co}^{2+}$  and  $\text{Cl}^-$  ions by changing the concentrations of either of the ions Viva Voce 5 Quantitative Analysis 1 To prepare M 10 oxalic acid solution by direct weighing method 2 To prepare M 10 solution of sodium carbonate by direct weighing method 3 To determine the strength of given solution of sodium hydroxide by titrating it against N 10 or M 20 solution of oxalic acid 4 To determine the strength of a given solution of hydrochloric acid by titrating it against a standard N 10 or M 20 sodium carbonate solution Viva Voce 6 Qualitative Analysis Analysis of Anions Analysis of Cations Viva Voce 7 Detection of Elements in Organic Compounds 1 To detect the presence of nitrogen sulphur and halogens in a given organic compound by Lassaigne's test 2 To detect the presence of nitrogen sulphur and halogens in the given organic compound sample number by Lassaigne's test Viva Voce INVESTIGATORY PROJECTS 1 Checking of Bacterial Contamination in Water 1 To check the bacterial contamination in drinking water by testing sulphide ions Viva Voce 2 Methods of Water Purification 1 To purify water from suspended impurities by using sedimentation 2 To purify water by boiling 3 To purify water by distillation method 4 To purify water by reverse osmosis technique 5 To purify water by GAC method 6 To purify water by bleach treatment 7 To purify water by oxidising agent 8 To purify water by ozone treatment method Viva Voce 3 Water Analysis 1 To test the hardness of different water samples Viva Voce 4 Foaming Capacity of Various Soaps 1 To compare the foaming capacity of different washing soaps 2 To study the effect of addition of sodium carbonate on foaming capacity of washing soap Viva Voce 5 Tea Analysis 1 To study the acidity of different samples of tea leaves tea by using pH paper Viva Voce 6 Analysis of Fruits and Vegetable Juices 1 To analysis the fruit and vegetable juices for the constituent present in them Viva Voce 7 Rate of Evaporation 1 To study the rate of evaporation of different liquids Viva Voce 8 Effect of Acids and Bases on Tensile Strength of Fibres 1 To compare the tensile strength of natural fibres and synthetic fibres 2 To study the effect of acids and bases on tensile strength of different fibres Viva Voce

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