

**CHEMISTRY (CODE - 043)**  
**SAMPLE QUESTION PAPER\***  
**CLASS XII (2025-26)**

**Time: 3 hours**

**Max. Marks: 70**

**GENERAL INSTRUCTIONS:**

**Read the following instructions carefully.**

1. There are **33** questions in this question paper with internal choice.
2. **SECTION A** consists of 16 multiple-choice questions carrying 1 mark each.
3. **SECTION B** consists of 5 short answer questions carrying 2 marks each.
4. **SECTION C** consists of 7 short answer questions carrying 3 marks each.
5. **SECTION D** consists of 2 case-based questions carrying 4 marks each.
6. **SECTION E** consists of 3 long answer questions carrying 5 marks each.
7. All questions are compulsory.
8. Use of log tables and calculators is not allowed.

<b>Section-A</b>		
Question 1 to 16 are multiple choice questions. Only one of the choices is correct. Select and write the correct choice as well as the answer to these questions.		
1	<p>Which of the following reaction will lead to formation of ethyl methyl ketone:</p> <p>A. heating <math>\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{OH}</math> with acidified <math>\text{Na}_2\text{Cr}_2\text{O}_7</math></p> <p>B. passing <math>\text{CH}_3\text{C}(\text{OH})(\text{CH}_3)</math> over heated copper</p> <p>C. ozonolysis of <math>\text{CH}_3\text{CH}_2\text{C}(\text{CH}_3)=\text{CHCH}_3</math></p> <p>D. acetylene on reaction with <math>\text{HgSO}_4/\text{H}_2\text{SO}_4</math></p>	1
2	<p>Consider the reaction and identify B and C</p> $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Cl} \xrightarrow{\text{NaOH, Ethanol}} \text{A}$ $\text{A} \xrightarrow{\text{H}_2\text{O, H}^+} \text{B} \quad \text{and} \quad \text{A} \xrightarrow{\text{O}_2/\text{B}_2\text{H}_6/\text{H}_2\text{O}_2/\text{OH}^-} \text{C}$ <p>A. B=C= Butanol</p> <p>B. B= Butanol, C=Butene</p> <p>C. B= Butan-2-ol, C= Butanol</p> <p>D. B= Butene, C=Butan-2-ol</p>	1
3	<p>The counter ion in the coordination compound <math>[\text{Co}(\text{NH}_3)_5(\text{NO}_2)] \text{Cl}_2</math> is</p> <p>A. Ammine</p> <p>B. Cobalt</p>	1

## Chemistry 2014 Code 043 Answers

**John Bonner, George William  
Curtis, Henry Mills Alden, Samuel  
Stillman Conant, Montgomery  
Schuyler, John Foord, Richard Harding  
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