

## II PUC MID-TERM EXAMINATION, OCTOBER - 2023

Time : 3 Hrs. 15 Mins.

SUBJECT : CHEMISTRY (30)

Max Marks : 70

### Instructions

- 1) The question paper has FIVE parts. All parts are compulsory.
- 2) a) Part-A carries 10 marks. Each question carries 1 mark.  
b) Part-B carries 16 marks. Each question carries 2 marks.  
c) Part-C carries 15 marks. Each question carries 3 marks.  
d) Part-D carries 20 marks. Each question carries 5 marks.  
e) Part-E carries 09 marks. Each question carries 3 marks.
- 3) In Part-A, questions first attempted answer will be considered for awarding marks.
- 4) Write balanced chemical equations and draw labeled diagrams, graphs wherever required.
- 5) Direct answers to the numerical problems without detailed steps and specific units for final answer will not carry any marks.
- 6) Use log tables and simple calculations if necessary. (Use of Scientific Calculator is not allowed)

### PART - A

1x15=15

- 1) Select the correct option from the given choices.
- 1) Solubility of a substance is depends on  
a) Nature of solute                      b) Temperature                      c) pressure                      d) All of these
- 2) Raw Mangoes shrivel into pickles when placed in the concentrated solution of NaCl due to  
a) Osmosis                      b) Reverse osmosis  
c) Osmotic pressure                      d) Lowering of vapour pressure
- 3) The charge required to liberate one gram equivalent of an element is  
a) 96,500 F                      b) 1 F                      c) 1C                      d) 10 F
- 4) Chemical composition of 'Rust' is  
a)  $\text{FeO} \cdot 2\text{H}_2\text{O}$                       b)  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$                       c)  $\text{Fe}_2\text{O}_3$                       d)  $\text{FeO}$
- 5) Unit of molar conductivity is  
a)  $\text{sm}^2 \text{mol}^{-1}$                       b)  $\text{scm}^{-1} \text{mol}^{-1}$                       c)  $\text{sm} \text{mol}^{-1}$                       d)  $\text{mol}^{-1} \text{m}^2$
- 6) In 3d series element which shows largest number of oxidation state is  
a) Cr                      b) Mn                      c) Fe                      d) V
- 7) With respect to actinoids, which of the following statement is not correct  
a) All are radioactive                      b) Diamagnetic in nature  
c) Highly active in the finely divided state                      d) Silvery metals
- 8) The ligand  $\text{SCN}^-$  is an example for  
a) Monodentate                      b) Didentate                      c) Tridentate                      d) Ambidentate
- 9) An example for heteroleptic complex is  
a)  $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]$                       b)  $[\text{CoCl}_2(\text{en})_2]$                       c)  $\text{Pt}(\text{NH}_3)_4\text{Cl}_2$                       d) All the above
- 10) In the reaction  $\text{R}-\text{Cl} + \text{KCN} \rightarrow \text{R}-\text{CN} + \text{KCl}$ . To obtain ethanenitrile  $\text{R}-\text{Cl}$  must be  
a) Chloro Ethane                      b) 1-Chloropropane                      c) Chloro methane                      d) 2-Chloro propane
- 11) The following reaction is known as  $\text{CH}_3\text{Br} + \text{NaI} \xrightarrow{\text{acetone}} \text{CH}_3\text{I} + \text{NaBr}$   
a) Sandmeyer's reaction                      b) Swart's reaction  
c) Friedel Craft's reaction                      d) Finkelstein reaction
- 12) The alcohol in which  $-\text{OH}$  group attached to  $\text{SP}^2$  hybridised carbon atom present next to the carbon-carbon double bond is  
a) Benzylic alcohol                      b) Vinylic alcohol                      c) Allylic alcohol                      d) None of the above
- 13) The IUPAC name of 'wood spirit' is  
a) Methanol                      b) Ethanol                      c) Methanal                      d) Ethanal

(11T03)

# Chemistry 2nd Paper Question

**Khattar Dinesh**



## Chemistry 2nd Paper Question:

*Jpsc Pt 1St & 2Nd Paper Question Bank 2024 | Answers With Detailed Explanation* Team Prabhat, 2024-05-20 This book is intended for those candidates who are preparing for the Jharkhand Public Service Commission JPSC preliminary examination. The book contains a selected compilation of questions related to the first and second papers including detailed explanations and answers in Hindi. The objective of this book is to explain the examination patterns and question types. It serves as a valuable resource for self assessment and improvement. Detailed explanations assist candidates in understanding the complexities of the exam providing them with support in comprehending the intricacies of the examination. **The**

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## Table of Contents Chemistry 2nd Paper Question

1. Understanding the eBook Chemistry 2nd Paper Question
  - The Rise of Digital Reading Chemistry 2nd Paper Question
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 2nd Paper Question
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry 2nd Paper Question
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 2nd Paper Question
  - Personalized Recommendations
  - Chemistry 2nd Paper Question User Reviews and Ratings
  - Chemistry 2nd Paper Question and Bestseller Lists
5. Accessing Chemistry 2nd Paper Question Free and Paid eBooks
  - Chemistry 2nd Paper Question Public Domain eBooks
  - Chemistry 2nd Paper Question eBook Subscription Services
  - Chemistry 2nd Paper Question Budget-Friendly Options
6. Navigating Chemistry 2nd Paper Question eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry 2nd Paper Question Compatibility with Devices
  - Chemistry 2nd Paper Question Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry 2nd Paper Question
  - Highlighting and Note-Taking Chemistry 2nd Paper Question
  - Interactive Elements Chemistry 2nd Paper Question
8. Staying Engaged with Chemistry 2nd Paper Question

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry 2nd Paper Question
9. Balancing eBooks and Physical Books Chemistry 2nd Paper Question
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry 2nd Paper Question
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 2nd Paper Question
- Setting Reading Goals Chemistry 2nd Paper Question
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 2nd Paper Question
- Fact-Checking eBook Content of Chemistry 2nd Paper Question
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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