

7-1 Review and Reinforcement

Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative
positive
crisscross

noble gas
electrons
anion

octet
ions
cation

electron configuration
polyatomic
empirical formula

1. An ionic bond is an attraction between oppositely charged ions.
2. Anions have a negative charge.
3. An atom becomes an ion by losing or gaining electrons.
4. The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
5. When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
6. The atoms that make up polyatomic ions are bonded together by covalent bonds.
7. The crisscross method can be used to write the formula for an ionic compound.
8. The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

losing

true

true

polyatomic

seven

true (released)

9. Calcium becomes a monatomic cation by gaining two electrons.

10. A cation has a positive charge.

11. A binary ionic compound contains only one kind of cation and one kind of anion.

12. Monatomic ions consist of more than one atom.

13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.

14. A great deal of energy is produced when an ionic compound is formed.

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Key word index United States. National Bureau of Standards. Technical Information and Publications Division,1978

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in Safety Assessment of Nanomaterials Henriqueta Louro,Maria João Silva,2022-05-18 Since its advent nanotechnologies are considered key enabling technologies that take advantage of a wide array of nanomaterials NMs for biomedical and industrial applications generating significant societal and economic benefits However such innovation increases human exposure to these substances through inhalation ingestion or dermal contact raising public health concerns Furthermore the NMs specific physicochemical properties that confer them unique beneficial characteristics can also elicit nano bio interactions leading to toxicity and concerns for public health In addition such properties can be affected by the surrounding matrix particularly when incorporated in complex matrices such as food products leading to secondary features potentially more relevant than primary characteristics for determining their toxicological outcome These nano specific issues raise the question of whether the NMs may produce adverse outcomes that are not accounted for when using conventional toxicological approaches to assess their safety Such uncertainties about the safety of NMs for human health and the environment may hamper a faster and more widespread exploration of their potentials In response the NMs definition has evolved and nanotoxicology has developed towards new and more integrative approach methods to support regulatory and policy actions This book provides a perspective on recent developments in the synthesis application and characterization of NMs and the related nanotechnologies focusing on nanotoxicology for their accurate safety assessment early in the product development stage The use of complex in vitro models including multicellular systems and organoids and omics based approaches such as transcriptomics or epigenomics have greatly contributed to an in depth understanding of the cellular and molecular mechanisms behind some NMs toxicity Such mechanistic knowledge is equally addressed in this book and has set the basis for a predictive nanotoxicology approach building on adverse outcome pathways In addition considering the knowledge provided by the above mentioned approaches insights into risk assessment standardization and regulation of NMs are also included Incorporating adequate nanosafety assessment early in the life cycle of NMs will allow the implementation of the safe and sustainable by design paradigm enabling safety to keep pace with innovation Chapters 10 and 15 are available open access under a Creative Commons Attribution 4 0 International License via link [springer.com](https://www.springer.com) **Electricity and Magnetism** ,1993 *Chemical Abstracts* ,1923 **Tests in Print** ,1999 **The Elementary School Library Collection** Lois Winkel,Eileen Palmer Burke,1990 *Cumulated Index Medicus* ,1981 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *California Science* ,2008 Science stimulates curiosity and student inquiry integrates powerful support for reading and science literacy reaches all learners through numerous components and

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