

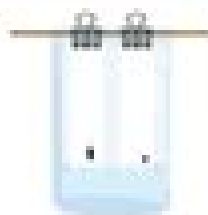
Define a pure substance

How can you distinguish a pure substance from an impure substance?

The  $T_m$  \_\_\_\_\_,  $T_b$  \_\_\_\_\_, and the  $M_r$  \_\_\_\_\_  $R_m$  \_\_\_\_\_

What will happen to the above if there are impurities in the sample?

What does chromatography separate?



Describe how the process works. Use the diagram to help separate solvent, paper, solute, mixture, spots,  $R_f$  value

Complete the word equation for calculating the  $R_f$  value

$R_f =$  \_\_\_\_\_

How does the  $R_f$  value allow you to identify a substance?

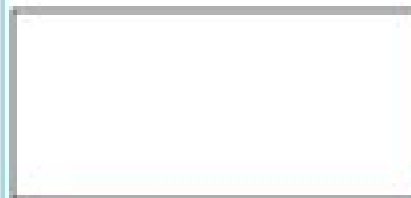
What colour does litmus go if chlorine is present?



What is the  $R_f$  value of the following chromatogram?

The distance moved by substance B is 30mm and the distance moved by solvent A is 120mm.

$$R_f = \frac{30}{120}$$



What are the 2 phases of chromatography?

$R_m$  \_\_\_\_\_ phase Describe this phase

$R_m$  \_\_\_\_\_ phase Describe this phase

Describe the test for oxygen.

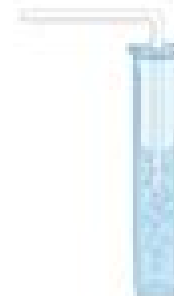
I understand the following topics...

I need to work on the following topics...

What gas does this experiment test for?



What gas does this experiment test for?



What colour does the litmus paper go if the gas is present?

# Chemistry 5 Self Check Activity

**Rachel S Tattersall**



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