

# Question 10 of 10 (100%)

1. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?  
 (a) 0.5  
 (b) 1  
 (c) 2  
 (d) 3

2. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

3. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

4. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

5. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

6. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

7. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

8. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

9. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

10. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

11. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

12. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

13. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

14. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

15. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

16. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

17. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

18. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

19. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

20. In  $\text{Fe}_2\text{O}_3 \cdot x\text{H}_2\text{O}$ ,  $x =$  ?

# Chemistry 8answer Key

**PT Brinkman**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the PT Brinkman text.

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