

**CHAPTER 14 REVIEW***Acids and Bases***SECTION 1****SHORT ANSWER** Answer the following questions in the space provided.

- 1.**
- Name the following compounds as acids:

|                                |                            |
|--------------------------------|----------------------------|
| _____ sulfuric acid _____      | a. $\text{H}_2\text{SO}_4$ |
| _____ sulfurous acid _____     | b. $\text{H}_2\text{SO}_3$ |
| _____ hydrosulfuric acid _____ | c. $\text{H}_2\text{S}$    |
| _____ perchloric acid _____    | d. $\text{HClO}_4$         |
| _____ hydrocyanic acid _____   | e. hydrogen cyanide        |

- 2.**
- \_\_\_\_\_
- $\text{H}_2\text{S}$
- Which (if any) of the acids mentioned in item
- 1**
- are binary acids?

- 3.**
- Write formulas for the following acids:

|                                      |                      |
|--------------------------------------|----------------------|
| _____ $\text{HNO}_2$ _____           | a. nitrous acid      |
| _____ $\text{HBr}$ _____             | b. hydrobromic acid  |
| _____ $\text{H}_3\text{PO}_4$ _____  | c. phosphoric acid   |
| _____ $\text{CH}_3\text{COOH}$ _____ | d. acetic acid       |
| _____ $\text{HClO}$ _____            | e. hypochlorous acid |

- 4.**
- Calcium selenate has the formula
- $\text{CaSeO}_4$
- .

|                                      |   |
|--------------------------------------|---|
| _____ $\text{H}_2\text{SeO}_4$ _____ | a. What is the formula for selenic acid?  |
| _____ $\text{H}_2\text{SeO}_3$ _____ | b. What is the formula for selenous acid? |

- 5.**
- Use an activity series to identify two metals that will not generate hydrogen gas when treated with an acid.

Choose from Cu, Ag, Au, Pt, Pd, or Hg.  
\_\_\_\_\_

- 6.**
- Write balanced chemical equations for the following reactions of acids and bases:

- a. aluminum metal with dilute nitric acid



- b. calcium hydroxide solution with acetic acid



# Describing Acids And Bases Section Review

**Fred M. Dewey**



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## **Table of Contents Describing Acids And Bases Section Review**

1. Understanding the eBook Describing Acids And Bases Section Review
  - The Rise of Digital Reading Describing Acids And Bases Section Review
  - Advantages of eBooks Over Traditional Books
2. Identifying Describing Acids And Bases Section Review
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Describing Acids And Bases Section Review
  - User-Friendly Interface
4. Exploring eBook Recommendations from Describing Acids And Bases Section Review
  - Personalized Recommendations
  - Describing Acids And Bases Section Review User Reviews and Ratings
  - Describing Acids And Bases Section Review and Bestseller Lists
5. Accessing Describing Acids And Bases Section Review Free and Paid eBooks
  - Describing Acids And Bases Section Review Public Domain eBooks
  - Describing Acids And Bases Section Review eBook Subscription Services
  - Describing Acids And Bases Section Review Budget-Friendly Options
6. Navigating Describing Acids And Bases Section Review eBook Formats

- ePub, PDF, MOBI, and More
- Describing Acids And Bases Section Review Compatibility with Devices
- Describing Acids And Bases Section Review Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Describing Acids And Bases Section Review
  - Highlighting and Note-Taking Describing Acids And Bases Section Review
  - Interactive Elements Describing Acids And Bases Section Review
- 8. Staying Engaged with Describing Acids And Bases Section Review
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Describing Acids And Bases Section Review
- 9. Balancing eBooks and Physical Books Describing Acids And Bases Section Review
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Describing Acids And Bases Section Review
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Describing Acids And Bases Section Review
  - Setting Reading Goals Describing Acids And Bases Section Review
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Describing Acids And Bases Section Review
  - Fact-Checking eBook Content of Describing Acids And Bases Section Review
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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