

## Types of Chemical Reactions Lab Answer Key

### Chemical Reaction #1

<i>Reactants (write in words)</i>	hydrochloric acid and magnesium
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	single replacement
<i>Predict Products (write in words)</i>	magnesium chloride and hydrogen gas
<i>Equation (write in symbols)</i>	$2\text{HCl}_{\text{(aq)}} + \text{Mg}_{\text{(s)}} \rightarrow \text{MgCl}_{2\text{(aq)}} + \text{H}_{2\text{(g)}}$

### Chemical Reaction #2

<i>Reactants (write in words)</i>	copper metal and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	synthesis
<i>Predict Products (write in words)</i>	copper oxide
<i>Equation (write in symbols)</i>	$2\text{Cu}_{\text{(s)}} + \text{O}_{2\text{(g)}} \rightarrow 2\text{CuO}_{\text{(s)}}$

### Chemical Reaction #3

<i>Reactants (write in words)</i>	methane gas and oxygen gas
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	combustion
<i>Predict Products (write in words)</i>	carbon dioxide and water
<i>Equation (write in symbols)</i>	$\text{CH}_{4\text{(g)}} + 2\text{O}_{2\text{(g)}} \rightarrow \text{CO}_{2\text{(g)}} + 2\text{H}_2\text{O}_{\text{(g)}}$

### Chemical Reaction #4

<i>Reactants (write in words)</i>	hydrogen peroxide
<i>Observations</i>	accept most responses
<i>Type of Reaction</i>	decomposition
<i>Predict Products (write in words)</i>	water and oxygen gas
<i>Equation (write in symbols)</i>	$2\text{H}_2\text{O}_{2\text{(l)}} \rightarrow 2\text{H}_2\text{O}_{\text{(l)}} + \text{O}_{2\text{(g)}}$

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**E Durkheim**



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