

Balancing Equations & Types of Reactions WS 1

Practice Problems

In your notebook, solve the following problems.

NOTE: A skeleton equation is an unbalanced equation.

SECTION 10.1 DESCRIBING CHEMICAL REACTIONS

1. Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.



2. Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.



3. Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.



4. Write a balanced equation for the production of HCl gas from its elements.



5. Write a sentence that completely describes the chemical reaction represented by this balanced equation.



Hydrochloric acid and calcium carbonate react to produce carbon dioxide, calcium chloride, and water.

6. Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.



2 solid silver reacts w/ 1 solid sulfur to create solid silver sulfide.

Describing Chemical Reactions Packet Answers

Ying-Ying Zheng



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Water Quality Engineering Mark M. Benjamin, Desmond F. Lawler, 2013-06-13 Explains the fundamental theory and mathematics of water and wastewater treatment processes By carefully explaining both the underlying theory and the underlying mathematics this text enables readers to fully grasp the fundamentals of physical and chemical treatment processes for water and wastewater Throughout the book the authors use detailed examples to illustrate real world challenges and their solutions including step by step mathematical calculations Each chapter ends with a set of problems that enable readers to put their knowledge into practice by developing and analyzing complex processes for the removal of soluble and particulate materials in order to ensure the safety of our water supplies Designed to give readers a deep understanding of how water treatment processes actually work Water Quality Engineering explores Application of mass balances in continuous flow systems enabling readers to understand and predict changes in water quality Processes for removing soluble contaminants from water including treatment of municipal and industrial wastes Processes for removing particulate materials from water Membrane processes to remove both soluble and particulate materials Following the discussion of mass balances in continuous flow systems in the first part of the book the authors explain and analyze water treatment processes in subsequent chapters by setting forth the relevant mass balance for the process reactor geometry and flow pattern under consideration With its many examples and problem sets Water Quality Engineering is recommended as a textbook for graduate courses in physical and chemical treatment processes for water and wastewater By drawing together the most recent research findings and industry practices this text is also recommended for professional environmental engineers in search of a contemporary perspective on water and wastewater treatment processes **Spotlight Science**

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