

## DETERMINING MOLECULAR FORMULAS (TRUE FORMULAS)

Solve the problems below

You may upload a picture of your work on paper, if you prefer...

1. The empirical formula of a compound is  $\text{NO}_2$ . Its molecular mass is 92 g/mol. What is its molecular formula?

2. The empirical formula of a compound is  $\text{CH}_2$ . Its molecular mass is 70 g/mol. What is its molecular formula?

3. A compound is found to be 40.0% carbon, 6.7% hydrogen, and 53.5% oxygen. Its molecular mass is 60 g/mol. What is its molecular formula?

4. A compound is 64.9% carbon, 13.5% hydrogen, and 21.6% oxygen. Its molecular mass is 74 g/mol. What is its molecular formula?

5. A compound is 54.5% carbon, 9.1% hydrogen, and 36.4% oxygen. Its molecular mass is 88 g/mol. What is its molecular formula?

# Determining Molecular Formulas True Formulas Answers

**Chemical Society (Great Britain)**



## **Determining Molecular Formulas True Formulas Answers:**

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In a world inundated with displays and the cacophony of quick connection, the profound power and psychological resonance of verbal artistry frequently disappear in to obscurity, eclipsed by the regular assault of sound and distractions. Yet, located within the musical pages of **Determining Molecular Formulas True Formulas Answers**, a fascinating function of fictional beauty that impulses with fresh emotions, lies an remarkable trip waiting to be embarked upon. Composed by a virtuoso wordsmith, that exciting opus courses visitors on a mental odyssey, gently revealing the latent possible and profound impact embedded within the complicated web of language. Within the heart-wrenching expanse of this evocative evaluation, we can embark upon an introspective exploration of the book is key themes, dissect its interesting writing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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