

UNIVERSITY OF EDUCATION
"Classroom" Semester-I, 2022
BSSE, Secondary, Session 2021-23

Course Code: EDUC2114
Subject: Classroom Assessment
Time Allowed: 100 Minutes.

Max. Marks: 56

Section II (Short Answer)

Q.2- Write short answers of the following.

3x8 = 24

- (i) Define three strategies to prevent "cheating" in exam?
- (ii) What is meant by objectivity of test?
- (iii) What are Rubrics? Enlist its types?
- (iv) What are supply types test item?
- (iv) State three purposes of Assessment?
- (v) Define reliability? Enlist its types?
- (vii) What is difference between evaluation and measurement?
- (viii) What is meant by achievement tests?

Section III (Essay Type)

Answer the following Questions

8x4 = 32

- Q.3:- Describe characteristics of good teacher in detail?
- Q.4:- Discuss Psychomotor Domain and write down its types?
- Q.5:- Define the concept of assessment for learning in assessment of learning in detail?
- Q.6:- Following are the scores obtained out of 150 Marks in the subject of Urdu compulsory of class 5th
1. Rank and order according to position (1st, 2nd and 3rd)
 2. Calculate in interpret range(50, 70, 73, 56, 66, 68, 85, 79, 80, 45, 120, 145, 135)

Department Of Education 2march Question Papers

Rosina Ehmann



Department Of Education 2march Question Papers:

School Education ,1896 **Votes & Proceedings** New South Wales. Parliament. Legislative Council,1871

Education in Thailand Gerald W. Fry,2018-11-01 This interdisciplinary book offers a critical analysis of Thai education and its evolution providing diverse perspectives and theoretical frameworks In the past five decades Thailand has seen impressive economic success and it is now a middle income country that provides development assistance to poorer countries However educational and social development have lagged considerably behind its globally recognized economic success This comprehensive book covers each level of education such as higher and vocational technical education and such topics as internationalization inequalities and disparities alternative education non formal and informal education multilingual education educational policy and planning and educational assessment The 25 Thai and 8 international contributors to the volume include well known academics and practitioners Thai education involves numerous paradoxes which are identified and explained While Thailand has impressively expanded its educational system quantitatively with much massification quality problems persist at all levels As such the final policy oriented summary chapter suggests strategies to enable Thailand to escape the middle income trap and enhance the quality of its education to ensure its long term developmental success

Parliamentary Papers Great Britain. Parliament. House of Commons,1904 School Government Chronicle and Education Authorities' Gazette ,1910 *Kenya Gazette* ,1953-01-20 The Kenya Gazette is an official publication of the government of the Republic of Kenya It contains notices of new legislation notices required to be published by law or policy as well as other announcements that are published for general public information It is published every week usually on Friday with occasional releases of special or supplementary editions within the week Kenya Gazette ,1953-02-03 The Kenya Gazette is an official publication of the government of the Republic of Kenya It contains notices of new legislation notices required to be published by law or policy as well as other announcements that are published for general public information It is published every week usually on Friday with occasional releases of special or supplementary editions within the week *Bookseller and the Stationery Trades' Journal* ,1913 **The Bookseller** ,1913 **Sessional Papers** Great Britain. Parliament. House of Commons,1901

Report of the Minister for Education and Cultural Activities Queensland. Department of Education,1917 *Minutes of evidence 16 February 1983, 23 February 1983, 2 March 1983, 9 March 1983, 10 March 1983* Northern Ireland. Assembly. Committee for Education,1983 Review of the Department of Education's policy on the rationalisation of schools concerning primary and secondary education in Northern Ireland

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.000731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web titration experiment in the neutralization of hydrochloric acid by sodium hydroxide the mole ratio of acid to base is 1 1
 $\text{HCl aq} + \text{NaOH aq} \rightarrow \text{NaCl aq} + \text{H}_2\text{O l}$
one mole of HCl would be fully neutralized by one mole of NaOH

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web conclusion the identification of an unknown solid acid the purpose of this experiment was to determine the standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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web acid base reactions can be used to determine concentrations of an acid or a base using stoichiometric calculations and titrimetric analysis the volume of a solution required to react

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web jul 19 2023 introduction titration is an analytical quantitative technique used to determine the concentration of a solute a pH titration is used to determine the concentration of an acid or a base titrations play an important role in determining amount and purity in many manufacturing processes

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web specifically an acid base titration can be used to figure out the concentration of an acid or base whether an unknown acid or base is strong or weak pK_a of an unknown acid or pK_b of the unknown base key terms titration a process where a solution of known strength is added to a certain volume of a treated sample containing an

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being NaOH reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4 + 2\text{NaOH aq} \rightarrow \text{Na}_2\text{C}_2\text{O}_4 + 2\text{H}_2\text{O}$ to differentiate an acid from a base in a solution a pH scale is used the pH scale ranges from 0 to 14 a pH of 7 is neutral in an acid base reaction

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web this point of titration is referred to as the equivalence point considering the 1 1 stoichiometry of this acid base reaction $\text{NaOH} + \text{C}_6\text{H}_4\text{COOH} \rightarrow \text{C}_6\text{H}_4\text{COONa} + \text{H}_2\text{O}$ the point of equivalence is the point of titration when the number of moles of NaOH added is equal to the number of moles of KHP in the

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