

## 11

## CHEMICAL REACTIONS

## SECTION 11.1 DESCRIBING CHEMICAL REACTIONS (pages 321–329)

*This section explains how to write equations describing chemical reactions using appropriate symbols. It also describes how to write balanced chemical equations when given the names or formulas of the reactants and products in a chemical reaction.*

## ► Writing Chemical Equations (pages 321–323)

1. A chemical reaction occurs when one or more \_\_\_\_\_ change into one or more new substances called \_\_\_\_\_.
2. The arrow in a reaction means \_\_\_\_\_.
3. Is the following sentence true or false? When there are two or more reactants or products, they are separated by an arrow. \_\_\_\_\_
4. Write a word equation that describes the following reactions.
  - a. Acetylene reacts with oxygen to produce carbon dioxide and water.  
\_\_\_\_\_
  - b. When heated, mercury(II) oxide reacts to form mercury and oxygen.  
\_\_\_\_\_
5. What is a chemical equation?  
\_\_\_\_\_  
\_\_\_\_\_
6. A chemical reaction that shows only the formulas, but not the relative amounts of the reactants and products is a(n) \_\_\_\_\_.
7. Identify the reactant(s) and product(s) in the chemical equation  $\text{Li} + \text{Br}_2 \longrightarrow \text{LiBr}$ .
  - a. reactant(s) \_\_\_\_\_
  - b. product(s) \_\_\_\_\_
8. Circle the letter of each statement that is true about a catalyst.
  - a. A catalyst is the new material produced as a result of a chemical reaction.
  - b. A catalyst is not used up in a chemical reaction.
  - c. A catalyst adds heat to a chemical reaction.
  - d. A catalyst speeds up a chemical reaction.

# Describing Chemical Change Answers

**Huangqi Zhang**



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## **Table of Contents Describing Chemical Change Answers**

1. Understanding the eBook Describing Chemical Change Answers
  - The Rise of Digital Reading Describing Chemical Change Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Describing Chemical Change Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Describing Chemical Change Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Describing Chemical Change Answers
  - Personalized Recommendations
  - Describing Chemical Change Answers User Reviews and Ratings
  - Describing Chemical Change Answers and Bestseller Lists
5. Accessing Describing Chemical Change Answers Free and Paid eBooks
  - Describing Chemical Change Answers Public Domain eBooks
  - Describing Chemical Change Answers eBook Subscription Services
  - Describing Chemical Change Answers Budget-Friendly Options

6. Navigating Describing Chemical Change Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Describing Chemical Change Answers Compatibility with Devices
  - Describing Chemical Change Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Describing Chemical Change Answers
  - Highlighting and Note-Taking Describing Chemical Change Answers
  - Interactive Elements Describing Chemical Change Answers
8. Staying Engaged with Describing Chemical Change Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Describing Chemical Change Answers
9. Balancing eBooks and Physical Books Describing Chemical Change Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Describing Chemical Change Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Describing Chemical Change Answers
  - Setting Reading Goals Describing Chemical Change Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Describing Chemical Change Answers
  - Fact-Checking eBook Content of Describing Chemical Change Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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