

**CHAPTER 11, Chemical Reactions** *(continued)*

9. Use the symbols in Table 11.1 on page 323 to write a skeleton equation for the following chemical reaction. Hydrochloric acid reacts with zinc to produce aqueous zinc(II) chloride and hydrogen gas.

**► Balancing Chemical Equations** (pages 324–328)

10. What is the law of conservation of mass?

In any physical or chemical change, mass is neither created nor destroyed.

11. Complete the flowchart for balancing equations.

Determine the correct formulas and physical states for the \_\_\_\_\_ **reactants** \_\_\_\_\_ and \_\_\_\_\_ **products** \_\_\_\_\_.

Write a \_\_\_\_\_ **skeleton equation** \_\_\_\_\_ with the formulas for the reactants on the left and the formulas for the products on the right of a yields sign ( $\longrightarrow$ ).

Count the number of \_\_\_\_\_ **atoms** \_\_\_\_\_ of each element in the reactants and in the products.

Balance the number of atoms of the elements on the two sides of the equation by placing \_\_\_\_\_ **coefficients** \_\_\_\_\_ in front of formulas. Never try to balance an equation by changing the \_\_\_\_\_ **subscripts** \_\_\_\_\_ in formulas.

Check each atom or polyatomic ion to be sure the equation is \_\_\_\_\_ **balanced** \_\_\_\_\_, and make sure that all coefficients are in the \_\_\_\_\_ **lowest** \_\_\_\_\_ possible ratio.

# Describing Chemical Reactions 1answer Key

**LM Reder**



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**Chemical Principles** Peter Atkins, Loretta Jones, 2007-08 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop chemical insight by showing the connections between fundamental chemical ideas and their applications Unlike other texts it begins with a detailed picture of the atom then builds toward chemistry's frontier continually demonstrating how to solve problems think about nature and matter and visualize chemical concepts as working chemists do Flexibility in level is crucial and is largely established through clearly labeling separating in boxes the calculus coverage in the text Instructors have the option of whether to incorporate calculus in the coverage of topics The multimedia integration of Chemical Principles is more deeply established than any other text for this course Through the unique eBook the comprehensive Chemistry Portal Living Graph icons that connect the text to the Web and a complete set of animations students can take full advantage of the wealth of resources available to them to help them learn and gain a deeper understanding Cehmistry Textbook for College and University USA Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA *Fundamentals of Environmental Chemistry, Third Edition* Stanley E. Manahan, 2011-03-05 Written by an expert using the same approach that made the previous two editions so successful Fundamentals of Environmental Chemistry Third Edition expands the scope of book to include the strongly emerging areas broadly described as sustainability science and technology including green chemistry and industrial ecology The new edition includes Increased emphasis on the applied aspects of environmental chemistry Hot topics such as global warming and biomass energy Integration of green chemistry and sustainability concepts throughout the text More and updated questions and answers including some that require Internet research Lecturers Pack on CD ROM with solutions manual PowerPoint presentations and chapter figures available upon qualifying course adoptions The book provides a basic course in chemical science including the fundamentals of organic chemistry and biochemistry The author uses real life examples from environmetnal chemistry green chemistry and related areas while maintaining brevity and simplicity in his explanation of concepts Building on this foundation the book covers environmental chemistry broadly defined to include sustainability aspects green chemistry industrial ecology and related areas These chapters are organized around the five environmental spheres the hydrosphere atmosphere geosphere biosphere and the anthrosphere The last two chapters discuss analytical chemistry and its relevance to environmental chemistry Manahan's clear concise and readable style makes the information accessible regardless of the readers level of chemistry knowledge He demystifies the material for those who need the basics of chemical science for their trade profession or study curriculum as well as for readers who want to have an understanding of the fundamentals of sustainable chemistry in its crucial role in maintaining a livable planet *Essentials of Medical Geology*, 2005-02-03 This authoritative reference volume emphasizes the importance and interrelationships of geological processes to the health and diseases of humans and animals Its accessible format fosters better communication between the health and geoscience communities by elucidating the

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