

Determine the empirical formula for each of the following, show all of your work.

1. The composition of an acid is 40.0% carbon, 6.71% hydrogen, and 53.29% oxygen, what is the empirical formula of this acid?
2. An unknown compound is analyzed and found to be composed of 14.79% nitrogen, 50.68% oxygen, and 34.53% zinc, calculate the empirical formula for the compound.
3. Calculate the empirical formula for a compound with the analysis of 74.97% aluminum and 25.03% carbon.
4. The composition of ascorbic acid (vitamin C) is 40.92% carbon, 4.58% hydrogen, and 54.50% oxygen. What is the empirical formula for vitamin C?
5. What is the empirical formula for a compound that is 26.56% potassium, 35.41% chromium, and 38.03% oxygen?
6. A new organic compound yielded the following chemical analysis, C = 60.59%, H = 7.12%. Further analysis revealed oxygen to be the only other element present. Compute the empirical formula for this compound.
7. Calculate the empirical formula for caffeine if the following percentages were determined, 49.5% C, 5.19% H, 28.9% N, and 16.5% O.

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