

# Assigning Oxidation Numbers

Rule	Example
1. The oxidation number of any uncombined element is 0.	The oxidation number of $\text{Na}(s)$ is 0.
2. The oxidation number of a monatomic ion equals the charge on the ion.	The oxidation number of $\text{Cl}^-$ is -1.
3. The more-electronegative element in a binary compound is assigned the number equal to the charge it would have if it were an ion.	The oxidation number of O in NO is -2.
4. The oxidation number of fluorine in a compound is always -1.	The oxidation number of F in LiF is -1.
5. Oxygen has an oxidation number of -2 unless it is combined with F, in which it is +1 or +2, or it is in a peroxide, in which it is -1.	The oxidation number of O in $\text{NO}_2$ is -2.
6. Hydrogen's oxidation state in most of its compounds is +1 unless it is combined with a metal, in which case it is -1.	The oxidation number of H in LiH is -1.
7. In compounds, the elements of Groups 1 and 2 as well as aluminum have oxidation numbers of +1, +2, and +3, respectively.	The oxidation number of Ca in $\text{CaCO}_3$ is +2.
8. The sum of the oxidation numbers of all atoms in a neutral compound is 0.	The oxidation number of C in $\text{CaCO}_3$ is +4.
9. The sum of the oxidation numbers of all atoms in a polyatomic ion equals the charge of the ion.	The oxidation number of P in $\text{H}_2\text{PO}_4^-$ is +5.

# Determining Oxidation Numbers Prentice Hall Key

**Fred H. Redmore**



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**Optimization of Dissolved Oxygen Measurements** H. C. Edgington,1971      **Instructors Resource Manual** John McMurry,Robert A. Pribush,Robert Fay,2000-12-27      **Basic Chemistry** William S. Seese,Guido H. Daub,1985

**Geoenvironmental Engineering** Hari D. Sharma,Krishna R. Reddy,2004-05-20 Geoenvironmental Engineering covers the application of basic geological and hydrological science including soil and rock mechanics and groundwater hydrology to any number of different environmental problems Includes end of chapter summaries design examples and worked out numerical problems and problem questions Offers thorough coverage of the role of geotechnical engineering in a wide variety of environmental issues Addresses such issues as remediation of in situ hazardous waste the monitoring and control of groundwater pollution and the creation and management of landfills and other above ground and in situ waste containment systems      **The Big Book of Herbs** Arthur O. Tucker,Thomas DeBaggio,2000 A typical entry provides the plant s botanical name and family whether it is an annual or perennial and its height hardiness light requirements water consumption and required soil type and pH The plant s name in various languages is included as is a history of the plant its chemistry how to propagate the plant and its culinary and landscape uses A botanical key is given to identify the plant and its description includes its country of origin and various data on the plants leaves flowers fruits and seeds      *New Technical Books* New York Public Library,1952      **Fundamentals of Chemistry** Fred H. Redmore,1979      **Chemical Properties of**

**Ground Water and Their Corrosion and Encrustation Effects on Wells** Ivan Barnes,Frank Eldridge Clarke,1969

**Geological Survey Professional Paper** ,1965      **The Software Encyclopedia 2001** ,2001      U.S. Geological Survey Professional Paper ,1964      **Geological Survey Professional Paper** Geological Survey (U.S.),1966      *The Electrical Engineer* ,1897      **Principles of Electronic Packaging** Donald P. Seraphim,Ronald Lasky,Che-Yu Li,1989      **Journal of the Society of Chemical Industry** Society of Chemical Industry (Great Britain),1890      *The Journal of the Society of Chemical Industry* ,1890      **Journal of the Society of Chemical Industry** ,1920      **Proceedings of the American Power Conference** ,1961      Analytical Chemistry Douglas A. Skoog,Donald M. West,1979      *Engineering* ,1928

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