

ঢাকা বোর্ড

১৪। সর্বাধিক তড়িৎ কদাচুৎকতায় মৌলের সর্বশেষ শক্তিস্তরের ইলেকট্রন বিন্যাস কোনটি?

- (ক) $5s^2 5p^5$
(খ) $4s^2 4p^5$
(গ) $3s^2 3p^5$
(ঘ) $2s^2 2p^5$

১৫। নিচের কোনটি তড়িৎ কদাচুৎকতায় সঠিক ক্রম?

- (ক) $Li > Na > K > Rb > Cs$
(খ) $Li > Na > K > Cs > Rb$
(গ) $Li > K > Na > Cs > Rb$
(ঘ) $Li > K > Na > Rb > Cs$

১৬। কোন শক্তিস্তরে ইলেকট্রন স্থানান্তরের জন্য হাইড্রোজেনের UV বর্ণালীরেখা পাওয়া যায়?

- (ক) $7 \rightarrow 1$
(খ) $6 \rightarrow 3$
(গ) $5 \rightarrow 2$
(ঘ) $4 \rightarrow 3$

১৭। নিচের কোন জোড়টি দিয়ে বাফার প্রকরণ তৈরি হয়?

- (ক) HNO_2 ও $NaNO_2$
(খ) HCl ও KCl
(গ) HNO_3 ও NH_4NO_3
(ঘ) $NaOH$ ও Na_2CO_3

১৮। ইলেকট্রন দ্বারা পূর্ণ হওয়ার জন্য অর্বিটালের কোন ক্রমটি সঠিক?

- (ক) $4s > 3p > 4p > 5s$
(খ) $4s > 3d > 4p > 5s$
(গ) $4s > 3d > 5p > 4d$
(ঘ) $5s > 4p > 5p > 4d$

১৯। $X(g) \rightarrow X(g)^+ + e^-$; বিক্রিয়াটি সংঘটিত হলে নিচের কোনটি প্রযোজন্য?

- (ক) আয়নীকরণ শক্তি
(খ) স্ব-প্রত্যায়ক
(গ) ইলেকট্রন আসক্তি
(ঘ) তড়িৎ কদাচুৎকতা

২০। AX_2 এর প্রাথমিক S হলে প্রাথমিক গণফল কত হবে?

- (ক) S^2
(খ) $4S^3$
(গ) $27S^4$
(ঘ) $108S^5$

নিচের উদ্বীপক বিক্রিয়া হতে ২১ ও ২২ নং প্রশ্নের উত্তর দাও :



২১। উদ্বীপকের বিক্রিয়াটিতে তাপের প্রভাব কীভাবে হবে?

- (ক) বাপ বাড়ালে উৎপাদ কমে
(খ) তাপ কমালে উৎপাদ বাড়ে
(গ) তাপের প্রভাব নেই
(ঘ) K_p কমে

২২। উদ্বীপকের বিক্রিয়ার ক্ষেত্রে K_p ও K_C এর মধ্যে সম্পর্ক কোনটি?

- (ক) $K_p = K_C (RT)^{-1}$
(খ) $K_C = K_p (RT)^{-1}$
(গ) $K_p = K_C$
(ঘ) $K_p = K_C (RT)^2$

২৩। গ্রীন কেমিস্ট্রিতে অধিক তাৎপর্যপূর্ণ নীতি কোনটি?

- (ক) প্রভাবকের ব্যবহার
(খ) নিরাপদ প্রাথমিক ব্যবহার
(গ) সর্বোত্তম এটিম ইকোনমি
(ঘ) দূষণের প্রতিরোধ

২৪। $0.05M$ H_2SO_4 এর pOH কত?

- (ক) 1
(খ) 1.30
(গ) 12.7
(ঘ) 13

২৫। $25^\circ C$ তাপমাত্রায় পানির আয়নিক গণফল 1×10^{-14} হলে $[H_3O^+]$ এর মান নিচের কোনটি?

- (ক) 10^{-14}
(খ) 10^{-7}
(গ) 10^7
(ঘ) 10^{14}

Dhaka Board Chemistry 2013 Hsc Board Question

Surendra Verma



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