

Chem 1020 Molarity worksheet, version 2

1. What is the molarity of a solution that contains 5.83 g LiCl in 1.50 L solution?

$$5.83 \text{ g LiCl} \times \frac{1 \text{ mol LiCl}}{42.389 \text{ g LiCl}} = 0.13754 \text{ mol LiCl}$$

$$M = \frac{\text{mol solute}}{\text{L solution}} = \frac{0.13754 \text{ mol LiCl}}{1.50 \text{ L solution}} = 0.0917 \text{ M}$$

2. What is the volume of a 0.300 M CuSO<sub>4</sub> solution that contains 12.4 g CuSO<sub>4</sub>?

$$12.4 \text{ g CuSO}_4 \times \frac{1 \text{ mol CuSO}_4}{159.6 \text{ g CuSO}_4} \times \frac{1 \text{ L solution}}{0.300 \text{ mol CuSO}_4} = 0.26 \text{ L solution}$$

3. What mass of solid KBr would be needed to prepare 750.0 mL of 1.75 M KBr solution? How specifically would this solution be prepared?

$$750.0 \text{ mL KBr soln} \times \frac{1 \text{ L soln}}{1000 \text{ mL soln}} \times \frac{1.75 \text{ mol KBr}}{1 \text{ L soln}} \times \frac{119.01 \text{ g KBr}}{1 \text{ mol soln}} = 156 \text{ g KBr}$$

Prepare this solution by measuring 156 g KBr into a 750 mL volumetric flask, and adding enough water to make 750 mL solution.

4. What is the % by mass of 28.3 g of solution which contains 3.56 g NaCl?

$$\% (\text{mass}) = \frac{\text{mass solute}}{\text{mass solution}} \times 100\% = \frac{3.56 \text{ g NaCl}}{28.3 \text{ g solution}} \times 100\% = 12.6\%$$

5. What volume of methanol is required to prepare 2.0 L of a 15.0% by volume solution?

$$\% (\text{vol}) = \frac{\text{vol solute}}{\text{vol solution}} \times 100\%$$

$$\text{vol solute} = \left( \frac{\% (\text{vol})}{100\%} \right) \times \text{vol solution} = \left( \frac{15.0\%}{100\%} \right) \times 2.0 \text{ L} = 0.30 \text{ L}$$

6. What volume of 12.0 M HCl is required to make 500.0 mL of 0.400 M HCl? How specifically would this solution be prepared?

$$M_1 V_1 = M_2 V_2$$

$$V_1 = \frac{M_2 V_2}{M_1} = \frac{(0.400 \text{ M})(500.0 \text{ mL})}{(12.0 \text{ M})} = 16.7 \text{ mL}$$

Prepare this solution by measuring 16.7 mL of 12.0 M HCl into a 500 mL volumetric flask, and adding water to make 500 mL of solution.

7. If 17.0 mL of 16.0 M H<sub>2</sub>SO<sub>4</sub> solution is diluted to 100.0 mL, what is the diluted concentration?

$$M_1 V_1 = M_2 V_2$$

$$M_2 = \frac{M_1 V_1}{V_2} = \frac{(16.0 \text{ M})(17.0 \text{ mL})}{(100.0 \text{ mL})} = 2.72 \text{ M}$$

# E2020 Chemistry Answer Key

**Bright Thinker**



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*Oceanographic Processes of Coral Reefs* Eric Wolanski, Michael J. Kingsford, 2024-02-29 In the last two decades since publication of the first edition substantial advancements have been made in the science the need for transdisciplinary approaches to coral reef protection greater than ever before This new edition now in full color throughout with accompanying animations goes beyond identifying foundational information and current problems to pinpoint science based solutions for managers stakeholders and policy makers Coral reefs are connected by currents that carry plankton and the larvae of many reef based organisms Further they supply food to reefs Currents also bring pollutants from the land and together with the atmosphere affect the surrounding ocean The chapters in this book provide a much needed review of the biophysics of reefs with an emphasis on the Great Barrier Reef as an ecosystem The focus is on interactions between currents waves sediment and the dynamics of coastal and reef based ecosystems The topographic complexity of reefs redirects mainstream currents creates tidal eddies mushroom jets boundary layers stagnation zones and this turbulence is enhanced by the oceanographic chaos in the adjoining Coral Sea This is the environment in which particles and organisms of a range of sizes live from tiny plankton to megafauna This generates faunal connectivity at scales of meters to thousands of km within the Great Barrier Reef and with the adjoining ocean Pollution from land use is increasing and remedial measures are described both on land and on coral cays The impact of climate change is quantified in case studies about mangroves and corals Modelling this biophysical complexity is increasing in sophistication and the authors suggest how the field can advance further

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