

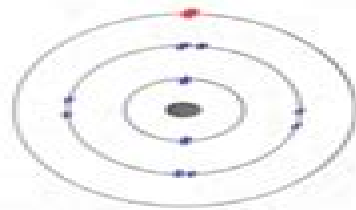
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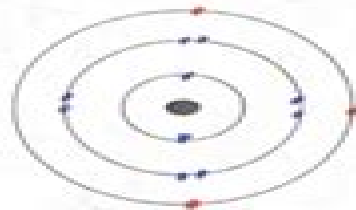
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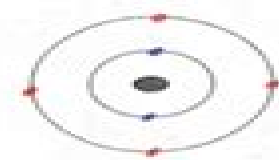
For each element draw the **inner electrons blue** & the **valence (outer) electrons red**.
The circles represent **possible** electron shells.



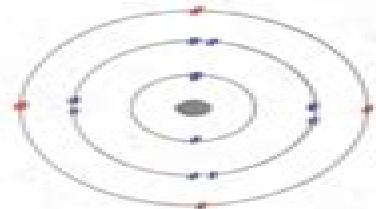
Sodium (Na) $\frac{11p^+}{11e^-}$



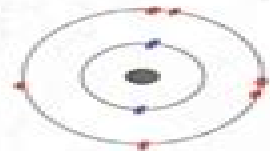
Aluminum (Al) $\frac{13p^+}{13e^-}$



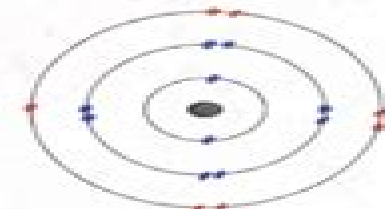
Carbon (C) $\frac{6p^+}{6e^-}$



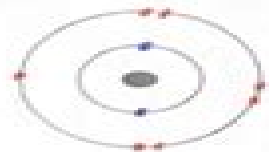
Silicon (Si) $\frac{14p^+}{14e^-}$



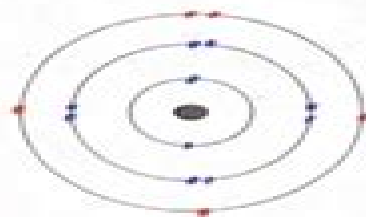
Oxygen (O) $\frac{8p^+}{8e^-}$



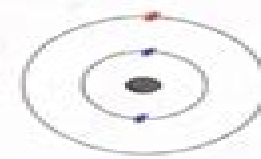
Chlorine (Cl) $\frac{17p^+}{17e^-}$



Fluorine (F) $\frac{9p^+}{9e^-}$



Phosphorus (P) $\frac{15p^+}{15e^-}$



Lithium (Li) $\frac{3p^+}{3e^-}$

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Eugene Maynard

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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