

Name Solutions

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### Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1 ve	transfers $1e^-$ to F	$Li^+$
F	7 ve	accepts $1e^-$ from Li	$F^-$

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) $Li + Cl \Rightarrow LiCl$ $Li + \cdot\ddot{Cl}: \rightarrow [Li]^+ [\ddot{Cl}:]^{-}$	Li	1 ve	donates $1e^-$	$Li^+$
	Cl	1 ve	accepts $1e^-$	$Cl^{-}$
2) $Ca + O \Rightarrow CaO$ $Ca + \cdot\ddot{O}: \rightarrow [Ca]^{+2} [\ddot{O}:]^{2-}$	Ca	2 ve	donates $2e^-$	$Ca^{+2}$
	O	6 ve	accepts $2e^-$	$O^{2-}$
3) $Be + F \Rightarrow BeF_2$ $Be + 2 \cdot\ddot{F}: \rightarrow [Be]^{+2} [2 \cdot\ddot{F}:]^{-}$	Be	2 ve	donates $2e^-$ to each F	$Be^{+2}$
	F	7 ve	each F accepts $1e^-$ each	$F^{-}$
4) $Mg + S \Rightarrow MgS$ $Mg + \cdot\ddot{S}: \rightarrow [Mg]^{+2} [\ddot{S}:]^{2-}$	Mg	2 ve	donates $2e^-$ to S	$Mg^{+2}$
	S	6 ve	accepts $2e^-$	$S^{2-}$
5) $K + F \Rightarrow KF$ $K + \cdot\ddot{F}: \rightarrow [K]^+ [\ddot{F}:]^{-}$	K	1 ve	donates $1e^-$	$K^{+1}$
	F	7 ve	accepts $1e^-$	$F^{-}$

# Bond Formation Study Guide Answers

**Susan McMurry**



## **Bond Formation Study Guide Answers:**

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. **Study Guide and Solutions Manual** Neil E. Schore, 2002-08-02

*Study Guide and Solutions Manual to Accompany Organic Chemistry, 11th Edition* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2013-03-25 This is the study guide and solutions manual to accompany Organic Chemistry 11th Edition. **Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition** Susan McMurry, 2000 Provides answers and explanations to all in-text and end-of-chapter exercises. Also includes summaries of name reactions, functional group synthesis, and reactions lists of reagents and abbreviations, and articles on topics ranging from infrared absorption frequencies to the Nobel Prize winners in Chemistry. This edition now includes all new artwork, expanded in-text problems, summary quizzes approximately every three chapters, more detailed explanations in solutions, and chapter outlines.

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*Study Guide/Selected Solutions Manual* Julie R. Frentrop, 2002-06 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all odd numbered end of chapter problems in the text book *Study Guide with Solutions to Accompany Burgoyne, A Study Course in Organic Chemistry* N. Sue Barnes, 1979

*Solutions Manual and Study Guide to Accompany Introduction to Organic Chemistry, 4th Ed* Paul A. Bartlett, Judith G. Koch, 1992 **Study Guide and Solutions Manual for Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1992

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**Study Guide with Selected Solutions** Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

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