

# CHEMICAL BONDING

## BASICS

### DOODLE NOTES

#### Bonding Basics

Small structure

Atoms which are directly joined are called **adjacent**.



**Adjacent-atom structure**

Atoms are the **adjacent** because they are **adjacent**.

Br

Three Main Types of Bonds

The type of bond is determined by the **electronegativity** difference between the two atoms.

**ionic bond**

Electronegativity is **high** difference between the two atoms. Electronegativity is **high** difference between the two atoms. Electronegativity is **high** difference between the two atoms.

Notes on:  
Three Main  
types of  
Chemical  
Bonds

**Covale Bond**

Formed between two **non-metal** atoms. The **electrons** are **shared** between the two atoms. Both atoms have the **same** electronegativity. Electrons are **shared** between the two atoms.

**Metallic Bond**

Only forms between **metal** atoms.

**Electronegativity difference between the two atoms is high.**

Covalent Bonds		Metallic and Ionic	
<b>Example</b> 	<b>Example</b> 	<b>Example</b> 	<b>Example</b> 
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# Bonding Basics Guided Notes

**John Calverley**



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