

Name \_\_\_\_\_ Date \_\_\_\_\_ Per \_\_\_\_\_

# WORKSHEET: Chemical Bonding – Ionic & Covalent!

## REMEMBER...

**Ionic Bond**

between a Metal and Non-Metal

(M + NM)

**Covalent Bond**

between a Non-Metal and Non-Metal

(NM + NM)

**PART 1:** Determine if the elements in the following compounds are metals or non-metals. Describe the type of bonding that occurs in the compound.

Compound	Element 1 (metal or non-metal?)	Element 2 (metal or non-metal?)	Bond Type
NO <sub>2</sub>	N = non-metal	O = non-metal	covalent
NaCl	Na = Metal (M)	Cl = Nonmetal (NM)	I
SO <sub>2</sub>	S = NM	O = NM	C
PI <sub>3</sub>	P = NM	I = NM	C
MgBr <sub>2</sub>	Mg = M	Br = NM	I
CaO	Ca = M	O = NM	I
H <sub>2</sub> O	H = NM	O = NM	C
K <sub>2</sub> O	K = M	O = NM	I
AlF <sub>3</sub>	Al = M	F = NM	I
O <sub>2</sub>	O = NM	O = NM	C
CuCl <sub>2</sub>	Cu = M	Cl = NM	I
NO <sub>2</sub>	N = NM	O = NM	C
CO <sub>2</sub>	C = NM	O = NM	C
HF	H = NM	F = NM	C
Rb <sub>2</sub> S	Rb = M	S = NM	I
NBr <sub>3</sub>	N = NM	Br = NM	C
Fe <sub>2</sub> O <sub>3</sub>	Fe = M	O = NM	I
CCl <sub>4</sub>	C = NM	Cl = NM	C

# Bonding Ionic Or Covalent Packet Answers

**Darrell D. Ebbing**



## **Bonding Ionic Or Covalent Packet Answers:**

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exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students

*Study Guide* Joan I. Senyk, Darrell D. Ebbing, 1984      **Study Guide for Chang's Chemistry** Kenneth W. Watkins, 1988      *Study Guide to Accompany Biology by Karen Arms and Pamela S. Camp* Russell C. Hollingsworth, 1979      **Study Guide and Solutions Manual to Accompany Fundamentals of Organic Chemistry** John McMurry, Susan McMurry, 1986      *Study Guide to Accompany Microbiology, Fourth Edition* George A. Wistreich, 1984      *Prentice Hall Chemistry*, 2000      *Exploring Earth and Space* Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter      **Study Guide for E. Russell Hardwick's Introduction to Chemistry** Joan Bouillon, 1984



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