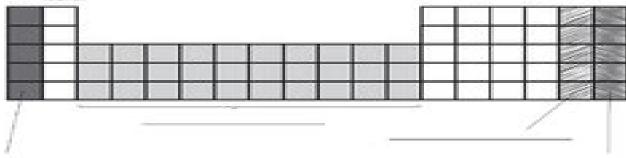
#### The Periodic Table

 Label the outline of the periodic lable below to show where these four sets of elements are found.



The periodic table is an arrangement of all known \_\_\_\_\_ organized according to their \_\_\_\_\_ properties. The elements are placed in order of increasing \_\_\_ number, which is the number of \_\_\_\_\_ in an atom of each element.

Each horizontal row in the periodic table is called a
\_\_\_\_\_. As you move from left to right across a period,
the atomic number \_\_\_\_\_\_, and elements generally
become \_\_\_\_\_\_ metallic in character. The periodic
trends within a period include changes in \_\_\_\_\_,
melting points, and boiling points.

The vertical columns of the periodic table are known as \_\_\_\_\_. Elements in the same group have similar \_\_\_\_\_ properties because they have the same number of \_\_\_\_\_ in their outermost shell. For example, Group 1 elements, also called the \_\_\_\_\_ metals, are very \_\_\_\_ and have \_\_\_\_\_ electron(s) in their outer shell. They react vigorously with \_\_\_\_ and become more reactive as you move \_\_\_\_\_ the group.

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