



Factors Affecting Solubility

- The amount of a substance that can dissolve in a particular solvent is referred to as its _____ in that solvent.
- Solubility is also affected by factors such as _____ and _____.
 - The solubility of solids in liquids generally _____ with temperature.
 - The solubility of gases in liquids generally _____ with temperature.
 - The solubility of _____ in liquids generally increases with pressure, whereas the solubility of _____ in liquids is unaffected by pressure.
- The more solute that is dissolved in a solvent, the more _____ there _____ the solution is.
 - A pale blue copper sulfate solution is _____ than a dark blue copper sulfate solution.

Saturated Solutions

- A solution that has the maximum amount of solute dissolved in it is called a _____ solution.
 - _____ a saturated solution containing a dissolved solid contains more solute to be dissolved.
 - _____ a solution containing a dissolved gas contains more solute to be dissolved.
- Solutions that contain more solute that can normally be dissolved are called _____ solutions.
- When a supersaturated solution containing a dissolved solid is cooled slowly, it can form _____.

unsaturated crystal solute gases heating pressure dissolving
saturated solid solvents supersaturated temperature



Cloze Activities 1 And 2 Solutions And Solubility Solutions

Hugh Westrup



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