

## Chemistry C1 Unit Opener

Picture Puzzler: Key Words	acid, thermometer, oxygen, melting The key word is atom.
Picture Puzzler: Close up	frost on a window

## C1 1.1 The particle model

In-text questions	A tiny particles B A material that has the same properties all the way through. C What its particles are like, how its particles are arranged, and how its particles move.
Activity	Vital vocab Students should offer explanations of the key words on this spread to a partner. Their definitions should be scientifically correct.
Summary questions	1 millions, particles, same, same, different, behaviour (6 marks) 2 $10\text{ cm}^3$ of mercury is heavier because its particles are heavier. (2 marks) 3 Visual summary should include key words, meanings of key words, examples to illustrate key words. Visual summary should be logically organised. (6 marks)

## C1 1.2 States of matter

In-text questions	A solid, liquid, gas B Solids cannot flow, liquids can flow. A Solids' shape is fixed, a liquid takes the shape of its container. Solids exist at colder temperatures than liquids. C You cannot compress a liquid because its particles touch their neighbours.
Activity	Express particle? The train travels at 0.135 km/s. This is $(0.135 \times 1000) = 135\text{ m/s}$ . So the oxygen particles travel faster.
Summary questions	1 There are three states of matter. You cannot compress a substance in the solid state because the particles touch each other. In the liquid and gas states, a substance flows because the particles can move from place to place. You can compress a gas because the particles are spread out. (4 marks) 2 A substance flows in the liquid state and in the gas state. You cannot compress a substance in the liquid state but you can compress it in the gas state. In the liquid state a substance takes the shape of the bottom of its container but in the gas state a substance takes the shape of the whole container. (3 marks) 3 QWC question (6 marks). Example answers: In the solid and liquid states the particles touch their neighbours so you cannot compress water in the solid and liquid states. In the gas state the particles do not touch their neighbours so you can compress water in the gas state. In the solid state the particles are in fixed positions so the shape of a solid is fixed. In the liquid and gas states particles move from place to place so the shape in the liquid and gas states is not fixed. In the liquid and gas states water can flow.

# Chapter 1 review Gases Section Answers

**Michael J. Padilla**



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