

Identifying Reaction Types & Balancing Equations

<u>Step 1:</u> Name the type of chemical reaction	<u>Step 2:</u> Balance the following chemical equations:
1. Ex: Synthesis	$__ \text{N}_2 + __ \text{H}_2 \rightarrow __ \text{NH}_3$
2.	$__ \text{KClO}_3 \rightarrow __ \text{KCl} + __ \text{O}_2$
3.	$__ \text{NaCl} + __ \text{F}_2 \rightarrow __ \text{NaF} + __ \text{Cl}_2$
4.	$__ \text{H}_2 + __ \text{O}_2 \rightarrow __ \text{H}_2\text{O}$
5.	$__ \text{AgNO}_3 + __ \text{MgCl}_2 \rightarrow __ \text{AgCl} + __ \text{Mg}(\text{NO}_3)_2$
6.	$__ \text{AlBr}_3 + __ \text{K}_2\text{SO}_4 \rightarrow __ \text{KBr} + __ \text{Al}_2(\text{SO}_4)_3$
7.	$__ \text{CH}_4 + __ \text{O}_2 \rightarrow __ \text{CO}_2 + __ \text{H}_2\text{O}$
8.	$__ \text{C}_3\text{H}_8 + __ \text{O}_2 \rightarrow __ \text{CO}_2 + __ \text{H}_2\text{O}$
9.	$__ \text{FeCl}_3 + __ \text{NaOH} \rightarrow __ \text{Fe}(\text{OH})_3 + __ \text{NaCl}$
10.	$__ \text{P} + __ \text{O}_2 \rightarrow __ \text{P}_2\text{O}_5$
11.	$__ \text{Na} + __ \text{H}_2\text{O} \rightarrow __ \text{NaOH} + __ \text{H}_2$
12.	$__ \text{Ag}_2\text{O} \rightarrow __ \text{Ag} + __ \text{O}_2$
13.	$__ \text{CO}_2 + __ \text{H}_2\text{O} \rightarrow __ \text{C}_6\text{H}_{12}\text{O}_6 + __ \text{O}_2$
14.	$__ \text{K} + __ \text{MgBr}_2 \rightarrow __ \text{KBr} + __ \text{Mg}$
15.	$__ \text{HNO}_3(\text{aq}) + __ \text{Ba}(\text{OH})_2(\text{aq}) \rightarrow __ \text{Ba}(\text{NO}_3)_2(\text{aq}) + __ \text{H}_2\text{O}(\text{l})$
16.	$__ \text{C}_5\text{H}_{12}(\text{g}) + __ \text{O}_2(\text{g}) \rightarrow __ \text{CO}_2(\text{g}) + __ \text{H}_2\text{O}(\text{g})$
17.	$__ \text{Al}(\text{s}) + __ \text{Fe}_2\text{O}_3(\text{aq}) \rightarrow __ \text{Al}_2\text{O}_3(\text{aq}) + __ \text{Fe}(\text{s})$
18.	$__ \text{Al}(\text{s}) + __ \text{O}_2(\text{g}) \rightarrow __ \text{Al}_2\text{O}_3(\text{s})$

Balancing Equations And Identifying Reaction Types

Answer Key

Laurie E. Westphal



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 - The Rise of Digital Reading Balancing Equations And Identifying Reaction Types Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Balancing Equations And Identifying Reaction Types Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Balancing Equations And Identifying Reaction Types Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Balancing Equations And Identifying Reaction Types Answer Key

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- Balancing Equations And Identifying Reaction Types Answer Key User Reviews and Ratings
- Balancing Equations And Identifying Reaction Types Answer Key and Bestseller Lists
- 5. Accessing Balancing Equations And Identifying Reaction Types Answer Key Free and Paid eBooks
 - Balancing Equations And Identifying Reaction Types Answer Key Public Domain eBooks
 - Balancing Equations And Identifying Reaction Types Answer Key eBook Subscription Services
 - Balancing Equations And Identifying Reaction Types Answer Key Budget-Friendly Options
- 6. Navigating Balancing Equations And Identifying Reaction Types Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Balancing Equations And Identifying Reaction Types Answer Key Compatibility with Devices
 - Balancing Equations And Identifying Reaction Types Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Balancing Equations And Identifying Reaction Types Answer Key
 - Highlighting and Note-Taking Balancing Equations And Identifying Reaction Types Answer Key
 - Interactive Elements Balancing Equations And Identifying Reaction Types Answer Key
- 8. Staying Engaged with Balancing Equations And Identifying Reaction Types Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Balancing Equations And Identifying Reaction Types Answer Key
- 9. Balancing eBooks and Physical Books Balancing Equations And Identifying Reaction Types Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Balancing Equations And Identifying Reaction Types Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Balancing Equations And Identifying Reaction Types Answer Key
 - Setting Reading Goals Balancing Equations And Identifying Reaction Types Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Balancing Equations And Identifying Reaction Types Answer Key

- Fact-Checking eBook Content of Balancing Equations And Identifying Reaction Types Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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