Net Ionic Equations

Select the correct balanced **net ionic equation** for each of the following reactions, occuring in <u>aqueous solution</u>.

A.
$$K_2CO_3 + Ca^{2+} \longrightarrow 2K^+ + CaCO_3$$

B.
$$Ca(NO_3)_2 + 2K^+ \longrightarrow Ca^{2+} + 2KNO_3$$

C.
$$(NH_4)_2SO_4 + 2Br^- \longrightarrow 2NH_4Br + SO_4^{2-}$$

D.
$$SO_4^{2-} + Ca^{2+} \longrightarrow CaSO_4$$

E.
$$2NH_4^+ + 2Br^- \longrightarrow 2NH_4Br$$

F.
$$2NO_3^- + 2K^+ \longrightarrow 2KNO_3$$

G.
$$SO_4^{2-} + 2Na^+ \longrightarrow Na_2SO_4$$

H.
$$2Cl^- + Cu^{2+} \longrightarrow CuCl_2$$

I.
$$CuSO_4 + Ca^{2+} \longrightarrow Cu^{2+} + CaSO_4$$

J.
$$CO_3^{2-} + Ca^{2+} \longrightarrow CaCO_3$$

K.
$$CaCl_2 + Cu^{2+} \longrightarrow Ca^{2+} + CuCl_2$$

Balancing Net Ionic Equations Answers

John Olmsted, Gregory M. Williams

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