# Key Answer

# More Balancing Equations KEY

```
2Au,S, + 15 H, - 4 Au + 6 H,S
3CaCl. + 2 Na.PO. → Ca.(PO.). + 6 NaCl
    4FeS + 7 O, → 2Fe,O, + 4SO,
       AlBr. + 3K -- 3KBr + Al
           P. + 6 Br. -- 4PBr.
    CaF. + LLSO, → CaSO, + 2 LIF
   Si(OH), * 4 Naftr -> Siftr, * 4NaOH
 JAgBr + GaPO. → Ag.PO. + GaBr.
            S, + 8 O, → 85O,
      6K + B.O. → 3K.O + 2B
           2N<sub>2</sub> + 5O<sub>2</sub> → 2 N<sub>2</sub>O<sub>4</sub>
        2 LI + 2 H,O -- 2 LIOH + H,
       5C + 2 SO, -- CS, + 4 CO
           4 Sb + 3 O, -- Sb,O,
       3 Ca + 2 AlCl, → 3 CaCl, + 2 Al
              450 + 5,--- 750.5.
          SiC + 2 Cl. -- SiCl. + C
         2AI + 3FeO -- AI.O. + 3Fe
             P. + 50. - 2P.O.
             3Mg + N, -- Mg,N,
         CJH, +5 O<sub>2</sub> -- 4H,O + 3 CO<sub>2</sub>
         Mg(ClO<sub>s</sub>), → MgCl<sub>s</sub> + 3 O<sub>s</sub>
             4 K + O. - 2 K.O
```

# **Balancing Practice Answer Key**

**Dheeraj Kumar Singh** 

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