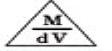


Density Calculations Worksheet I



Find the unknown quantity:

a)	d = 3 g/mL
	V = 100 mL
	M = ?
	$m = d \times v$
	= (3glmL)(100mL)
	= 3 × 10 2 3

0) d = ?
V = 950 mL
M = 95 g
d =
$$\frac{m}{V}$$

= $\frac{959}{750mL}$ = 0.10g/mL

= ?
= 950 mL
I = 95 g
=
$$\frac{M}{V}$$

= $\frac{959}{750mL}$ = 0.109 mL
C) d = 0.5 g/cm³
V = ?
M = 20 g
V = $\frac{M}{V}$
= $\frac{209}{0.59 \text{ cm}^3}$ = 4× 10 cm³

Find the unknown quantity (CONVERT FIRST to g or mL)

WORD PROBLEMS

A block of aluminum occupies a volume of 15.0 mL and weighs 40.5 g. What is its density?

Mercury metal is poured into a graduated cylinder that holds exactly 22.5 mL. The mercury used to fill the cylinder weighs 306.0 g. From this information, calculate the density of mercury.

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Ensheng Dong

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