#### **DENSITY WORKBOOK**

Question: What is Density?

Answer: <u>Density</u> is a measure of how much mass is contained in a given unit volume. Put simply, if mass is a measure of how much 'stuff' there is in an object; density is a measure of how tightly that 'stuff' is packed together. The principle of density was discovered by the Greek scientist Archimedes.

## Overview of Donsity Workshoot

Key formulas/Concepts:

 Density - Density = Mass divided by Volume (D = M/V). The mass of an object is 25 grams. The volume of an object is 5 cm<sup>3</sup>. D = 25g/5cm<sup>3</sup> = 5 g/cm<sup>3</sup>.

 Volume of a rectangular shaped object/cube - Length X Width X Height (L X W X H).

Example - A cube has a length of 3 cm. (Note - a cube has six equal sides).  $V = 3 \times 3 \times 3 = 27$  cm<sup>3</sup>. A piece of wood has a length of 10 cm, width of 18 cm, and height of 3 cm.  $V = 10 \times 18 \times 3 = 540$  cm<sup>3</sup>.

 Volume of an irregularly shaped object - Use a graduated cylinder filled with a fluid. Record the beginning amount of fluid (in mL - milliliters). Drop the object into the graduated cylinder. Record the level of the fluid with the object (in mL). To determine the volume: Level of fluid with object (minus) beginning amount of fluid. Example - a graduated cylinder is filled to the 30mL mark with water. You drop in a rock. The water level rises to 48mL. V = 48mL - 30mL = 18cm<sup>3</sup>.

- ✓ A submerged object displaces a volume of liquid equal to the volume of the object.
- ✓ One milliliter (1 mL) of water has a volume of 1 cubic centimeter (1cm³).

Problems (Always Show Your Work and units - g, mL, cm3, g/cm3, g/mL):

\_\_\_\_\_3. You have a box that has a volume of 210 cm<sup>3</sup> and mass of 42g.

What is its density?

# **Density Workbook Answer Guide**

Jane B. Taylor, John D. Jackson

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