


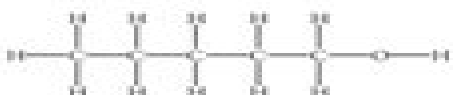

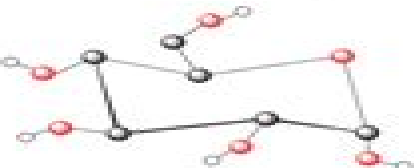
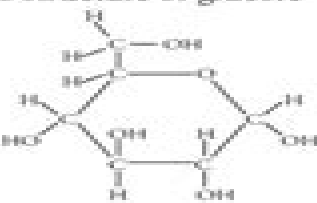
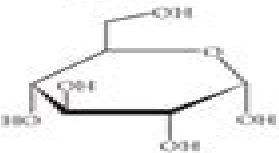
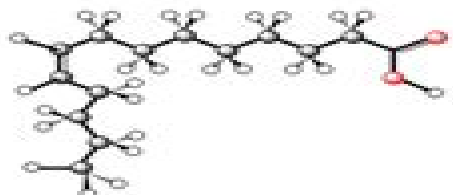


Biochemistry Basics

What concepts from chemistry are helpful in studying biology?

Why?

Typically chemistry is a prerequisite course for advanced biology courses. This is because everything in your body, everything in a plant, everything in a virus, etc. is made of atoms. The structures and properties of the molecules in an organism determine the features and properties of the organism. Which molecules are polar, which are nonpolar? Which molecules have acidic properties, which have basic properties? A quick review of these concepts at the beginning of your advanced biology course will help you to understand the molecular basis for life.

Model 1 – Molecular Drawings

<p>Ball-and-stick model of 1-pentanol</p>  A ball-and-stick model of 1-pentanol. Carbon atoms are represented by black spheres, hydrogen atoms by white spheres, and the oxygen atom by a red sphere. The molecule shows a five-carbon chain with a hydroxyl group at the end.	<p>Lewis structure of 1-pentanol</p>  A Lewis structure of 1-pentanol. It shows a horizontal chain of five carbon atoms, each with its valence electrons explicitly drawn. The terminal carbons are bonded to hydrogen atoms, and the rightmost carbon is bonded to an oxygen atom, which is in turn bonded to a hydrogen atom.	<p>Line drawing of 1-pentanol</p>  A line drawing of 1-pentanol. It consists of a zigzag line representing the carbon chain, with a short line segment at the end connected to an 'OH' group.
<p>Ball-and-stick model of glucose</p>  A ball-and-stick model of glucose in its cyclic pyranose form. Carbon atoms are black, hydrogen atoms are white, and oxygen atoms are red. The structure shows the six-membered ring and the attached hydroxyl groups.	<p>Lewis structure of glucose</p>  A Lewis structure of glucose in its cyclic form. It shows the six-membered ring with all atoms and their valence electrons explicitly drawn, including the hydroxyl groups on each carbon.	<p>Line drawing of glucose</p>  A line drawing of glucose in its cyclic form. It shows a hexagon representing the ring, with an oxygen atom at one vertex and hydroxyl groups attached to each carbon atom in a specific orientation.
<p>Ball-and-stick model of unsaturated fatty acid</p>  A ball-and-stick model of an unsaturated fatty acid. It features a long hydrocarbon chain with a distinct bend at the site of the double bond, followed by a carboxyl group at the end.	<p>Lewis structure of unsaturated fatty acid</p>  A Lewis structure of an unsaturated fatty acid. It shows a long chain of carbon atoms with a double bond, and a carboxyl group at the end, with all valence electrons explicitly drawn.	<p>Line drawing of unsaturated fatty acid</p>  A line drawing of an unsaturated fatty acid. It shows a zigzag line for the carbon chain with a sharp kink at the double bond, ending in a carboxyl group represented by a double bond to an oxygen and a single bond to an 'OH'.

1. Name the three molecules that are illustrated in Model 1.
2. Name the three types of drawings that are used to illustrate the molecules in Model 1.

Biochemistry Basics Pogil Key

Milin Kurup



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