

Molecular Compounds: Names and Formulas Worksheet

1. Write the formulas for the following compounds.

- (a) carbon dioxide CO_2
 (b) silicon dioxide SiO_2
 (c) water H_2O
 (d) carbon disulfide CS_2
 (e) sulfur trioxide SO_3
 (f) ammonia NH_3
 (g) carbon tetrachloride CCl_4
 (h) hydrogen peroxide H_2O_2
 (i) methane CH_4
 (j) ozone (trioxygen) O_3

- (k) diphosphorus trioxide P_2O_3
 (l) nitrogen monoxide NO
 (m) chlorine dioxide ClO_2
 (n) dinitrogen oxide N_2O
 (o) carbon monoxide CO
 (p) arsenic tribromide AsBr_3
 (q) phosphorus pentabromide PBr_5
 (r) dinitrogen tetroxide N_2O_4
 (s) silicon carbide SiC
 (t) sulfur dioxide SO_2

2. Write the names for the following compounds.

- (a) CF_4 carbon tetrafluoride
 (b) NH_3 ammonia
 (c) PBr_3 phosphorus tribromide
 (d) O_3 ozone
 (e) F_2 (gas) fluorine gas
 (f) CS_2 carbon disulfide
 (g) N_2O_4 dinitrogen tetroxide
 (h) H_2O_2 hydrogen peroxide
 (i) CO carbon monoxide
 (j) SiC silicon carbide

- (k) P_2O_5 diphosphorus pentoxide
 (l) CH_4 methane
 (m) SO_3 sulfur trioxide
 (n) H_2O water
 (o) SiO_2 silicon dioxide
 (p) PCl_5 phosphorus pentachloride
 (q) I_2 (gas) iodine gas
 (r) NO_2 nitrogen dioxide
 (s) SF_6 sulfur hexafluoride
 (t) H_2 (gas) hydrogen gas

Blackline Master 5 8

William Hazzard Barnwell



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