

Chapter 2

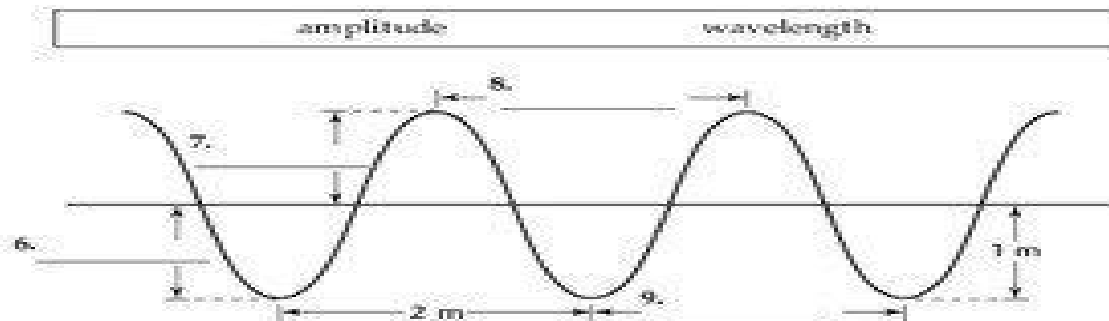
Use with Section 2

REINFORCEMENT**• Wave Properties**

Complete each sentence by circling the correct answers and crossing out the incorrect answers.

1. The amplitude of a wave can be measured from the (medium, crest) or the (trough, wavelength) to the rest position of the wave's medium.
2. Waves with greater amplitudes carry (more, less) energy than waves with smaller amplitudes.
3. The wavelength of a transverse wave is often measured from (crest to crest, crest to trough).
4. The number of waves that pass a point in one (second, minute) is the wave's (amplitude, frequency).
5. Waves with longer wavelengths have a (lower, higher) frequency and waves with shorter wavelengths have a (lower, higher) frequency.

Use the words in the box to label the diagram. You will use each term more than once. Then answer the questions.



10. What is the wavelength of the wave shown in the diagram? _____
11. What is the amplitude of the wave shown in the diagram? _____

Calculating Wave Properties Answers

James T. Shipman, Jerry D. Wilson



Calculating Wave Properties Answers:

The Fluid Earth Teacher's Guide , **Survival Guide to General Chemistry** Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **Princeton Lectures** Princeton University, 1920 **Shipman Phys Sci 6e Study Guide** James T. Shipman, Jerry D. Wilson, 1990

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