

# + WS 7.1 Specific Heat & Calorimetry

Show all work neatly.....

$$q = m c \Delta T$$

Specific Heats	
substance	c (J/g°C)
water	4.184
ethanol	2.452
graphite	0.720
diamond	0.502
iron	0.444
copper	0.385
silver	0.237
gold	0.129
ice	2.092

1. How much heat is required to raise the temp of 654 g of water from 34.5°C to 89.7°C?

Ans \_\_\_\_\_

2. How much heat is required to raise the temp of 654 g of silver from 34.5°C to 89.7°C?

Ans \_\_\_\_\_

3. If 7350 J were added to 152 g of ethanol, its temp would go up by how much?

Ans \_\_\_\_\_

4. 16.25 g of water at 54.0°C releases 402.7 J. What will be its final temp?

hint: it's cooling down, so the final temperature will be less than 54.0°C

Ans \_\_\_\_\_

5. 697 J are added to a 36.8 g of kerosene and the temp increases from 22.5°C to 34.7°C. Determine kerosene's specific heat.

Ans \_\_\_\_\_

6. 25 copper pennies (each weighing 3.12 g) are placed in 36.0 g of ethanol at room temp (22.1°C). How much heat will it take to raise the temperature up to 65.8°C?

hint: Calculate q for the copper & q for the ethanol separately. Then add your 2 answers together

Ans \_\_\_\_\_

7. What mass of 54.0°C water must be added to 468 g of 21.0°C water to make the final temp of both come out to be 29.0°C?

Ans \_\_\_\_\_

8. What mass of 54.0°C gold must be added to 468 g of 21.0°C water to make the final temp of both come out to be 29.0°C?

Ans \_\_\_\_\_

9. A 325 g brass rod at 100.0°C is placed in a cup containing 162 g of 24.3°C water. The final temp comes out to be 37.4°C. Determine brass's specific heat.

Ans \_\_\_\_\_

10. 100.0 g of water at 20.0°C are mixed with 200.0 g of copper at 40.0°C. What will the final temp come out to be?

Ans \_\_\_\_\_

# Calculations Involving Specific Heat Answer Key

**Jicheng Xie**



## **Calculations Involving Specific Heat Answer Key:**

## Unveiling the Energy of Verbal Beauty: An Mental Sojourn through **Calculations Involving Specific Heat Answer Key**

In a world inundated with displays and the cacophony of immediate connection, the profound energy and emotional resonance of verbal artistry usually disappear in to obscurity, eclipsed by the regular barrage of noise and distractions. However, located within the musical pages of **Calculations Involving Specific Heat Answer Key**, a interesting work of fictional elegance that impulses with raw feelings, lies an unforgettable trip waiting to be embarked upon. Published by way of a virtuoso wordsmith, that exciting opus instructions viewers on a psychological odyssey, gently revealing the latent possible and profound affect stuck within the elaborate internet of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is main themes, dissect their charming publishing style, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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### **Table of Contents Calculations Involving Specific Heat Answer Key**

1. Understanding the eBook Calculations Involving Specific Heat Answer Key
  - The Rise of Digital Reading Calculations Involving Specific Heat Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Calculations Involving Specific Heat Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Calculations Involving Specific Heat Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Calculations Involving Specific Heat Answer Key
  - Personalized Recommendations

- Calculations Involving Specific Heat Answer Key User Reviews and Ratings
- Calculations Involving Specific Heat Answer Key and Bestseller Lists
- 5. Accessing Calculations Involving Specific Heat Answer Key Free and Paid eBooks
  - Calculations Involving Specific Heat Answer Key Public Domain eBooks
  - Calculations Involving Specific Heat Answer Key eBook Subscription Services
  - Calculations Involving Specific Heat Answer Key Budget-Friendly Options
- 6. Navigating Calculations Involving Specific Heat Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Calculations Involving Specific Heat Answer Key Compatibility with Devices
  - Calculations Involving Specific Heat Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Calculations Involving Specific Heat Answer Key
  - Highlighting and Note-Taking Calculations Involving Specific Heat Answer Key
  - Interactive Elements Calculations Involving Specific Heat Answer Key
- 8. Staying Engaged with Calculations Involving Specific Heat Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Calculations Involving Specific Heat Answer Key
- 9. Balancing eBooks and Physical Books Calculations Involving Specific Heat Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Calculations Involving Specific Heat Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Calculations Involving Specific Heat Answer Key
  - Setting Reading Goals Calculations Involving Specific Heat Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Calculations Involving Specific Heat Answer Key
  - Fact-Checking eBook Content of Calculations Involving Specific Heat Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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