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## Chapter 6

Classifying Chemical Reactions

1. Balance each equation.
2. Classify the following reactions as synthesis, decomposition, combustion, single replacement, or double replacement.

Type of Reaction

decomposition



double replacement



combustion



single replacement



single replacement



synthesis &amp; combustion



combustion

# Blank Classifying Chemical Reactions

**Spencer L. Seager, Michael R.  
Slabaugh**



## Blank Classifying Chemical Reactions:

**Manual of Classification** United States. Patent and Trademark Office,1990 Includes list of replacement pages

**Spectrophotometric Determination of Urea in Natural Waters with Hypochlorite and Phenol** Robert T.

Emmet,1969 A yellow compound formed from urea hypochlorite and phenol with an absorbancy maximum at 454 millimicrons has been adapted to determine micromolar quantities of urea in natural waters The urea is chlorinated in a solution of sodium hypochlorite at pH 7.7 and upon the addition of phenol the intermediate condenses rapidly to form the colored product In freshwater the absorbancy index is 2600 absorbancy units per centimeter cell times gram atom urea N per liter and in 35 parts per thousand seawater the color is twice as intense The lower limit of detection of urea in seawater is 0.2 microgram atoms urea N per liter and the relative standard deviation is 10% at a urea N concentration of 1 microgram atom per liter An adaptation of the method to automatic analysis of clinical samples is suggested and a sensitive analysis procedure for tyrosine is described Safety-Scale Lab Exp Biochem 2e Spencer L. Seager,Michael R. Slabaugh,1994-05

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**Mathematical Teaching and Its Modern Methods** Truman Henry Safford,1886 *Official Gazette of the United States Patent Office* USA Patent Office,1920 An Elementary Course in Practical Zoology Buel Preston Colton,1890

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