

L1: Chapter 2 Identifying polar molecules

1. Look at the general structure of each molecule. 2. Comment on what functional groups are present and what properties they will have.

Complete the following table for the commonest (important biological) functional groups:

Use names: Figure 2.19 for help

Functional Group	Structure (showing H)	Hydrogen Bond Donor?	Hydrogen Bond Acceptor?	Properties
Alcohol	$\text{R}-\text{OH}$			Most important, responsible for solubility, usually non-polarised
Carboxyl	$\text{R}-\text{COOH}$			Acidic, often soluble
Carbonyl	$\text{R}-\text{C}(=\text{O})-\text{OH}$			Acidic, often soluble
Amino	$\text{R}-\text{NH}_2$			Basic, soluble
Sulphydryl	$\text{R}-\text{SH}$			Hydrophobic, important in protein structure
Phosphate	$\text{R}-\text{O}-\text{P}(=\text{O})(\text{OH})_2$			Acidic, soluble, for 1st time appeared in

Chapter 4 Reading Guide Chemistry

Terry C. Jones



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